The background of the slide is a photograph of a research vessel on a large body of water, likely Lake Superior. A large black crane arm extends from the vessel, and a white net is being lowered into the water. A person wearing a red jacket and a blue hard hat is visible on the right side of the vessel, looking towards the net. The water is dark blue with some whitecaps, and the sky is a pale blue.

# Why is Lake Superior so susceptible to methylmercury accumulation?

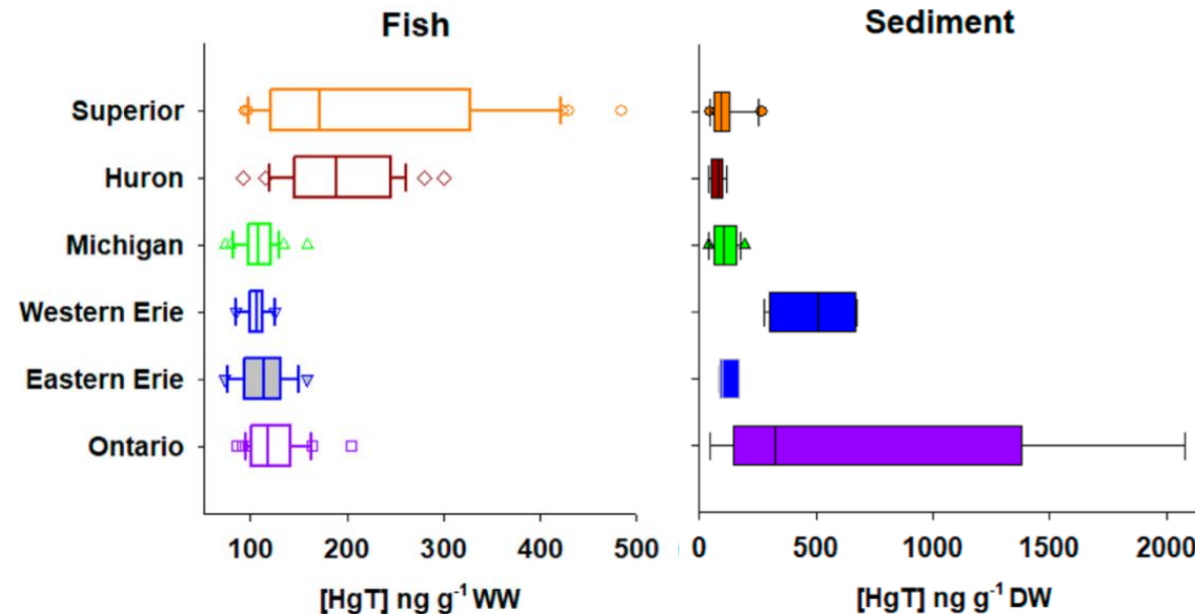
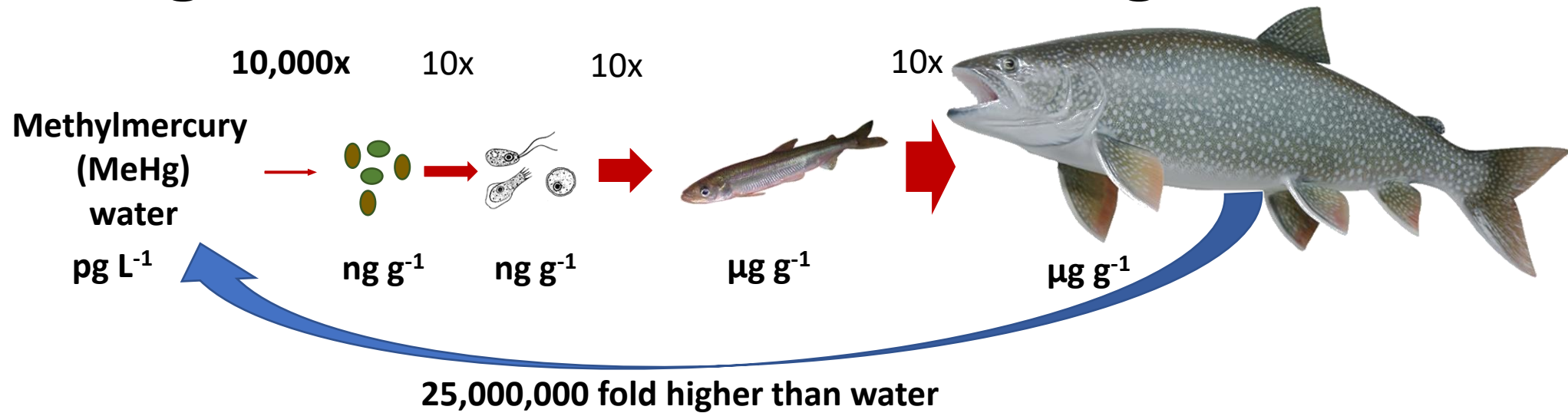
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*Ryan F. Lepak<sup>a,b</sup>, Jacob M Ogorek<sup>b</sup>, Joel Hoffman<sup>a</sup>, John F DeWild<sup>b</sup>, Michael T. Tate<sup>b</sup>, and David P. Krabbenhoft<sup>b</sup>*

*<sup>a</sup>USEPA GLTED*

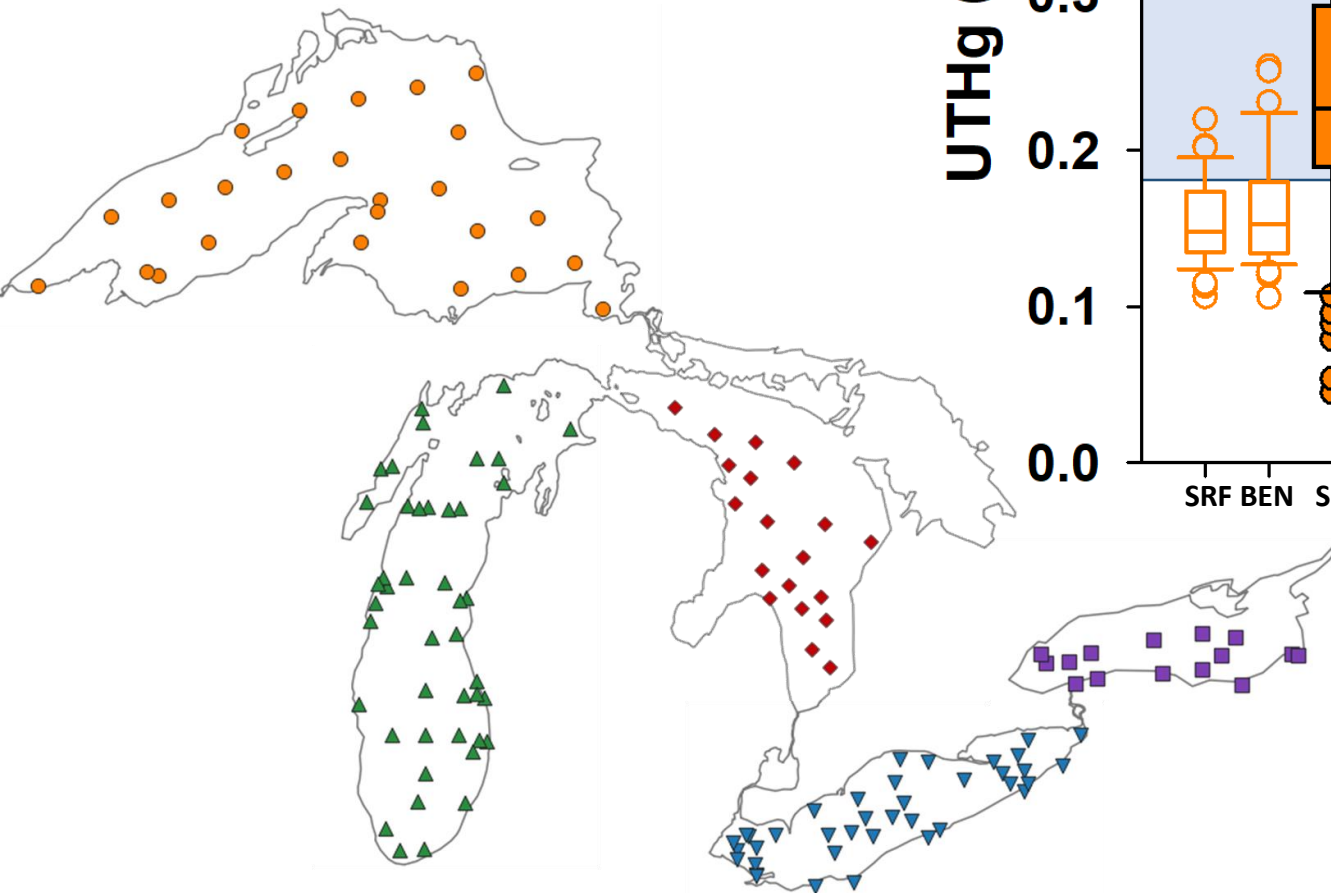
*<sup>b</sup>USGS Hg Research Lab*

# Hg bioaccumulation and magnification



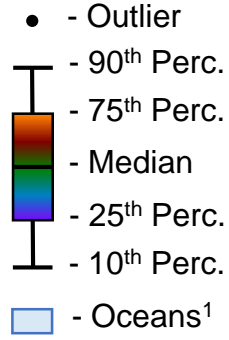
# Total Hg in water

- 143 sample locations
- 511 profiles

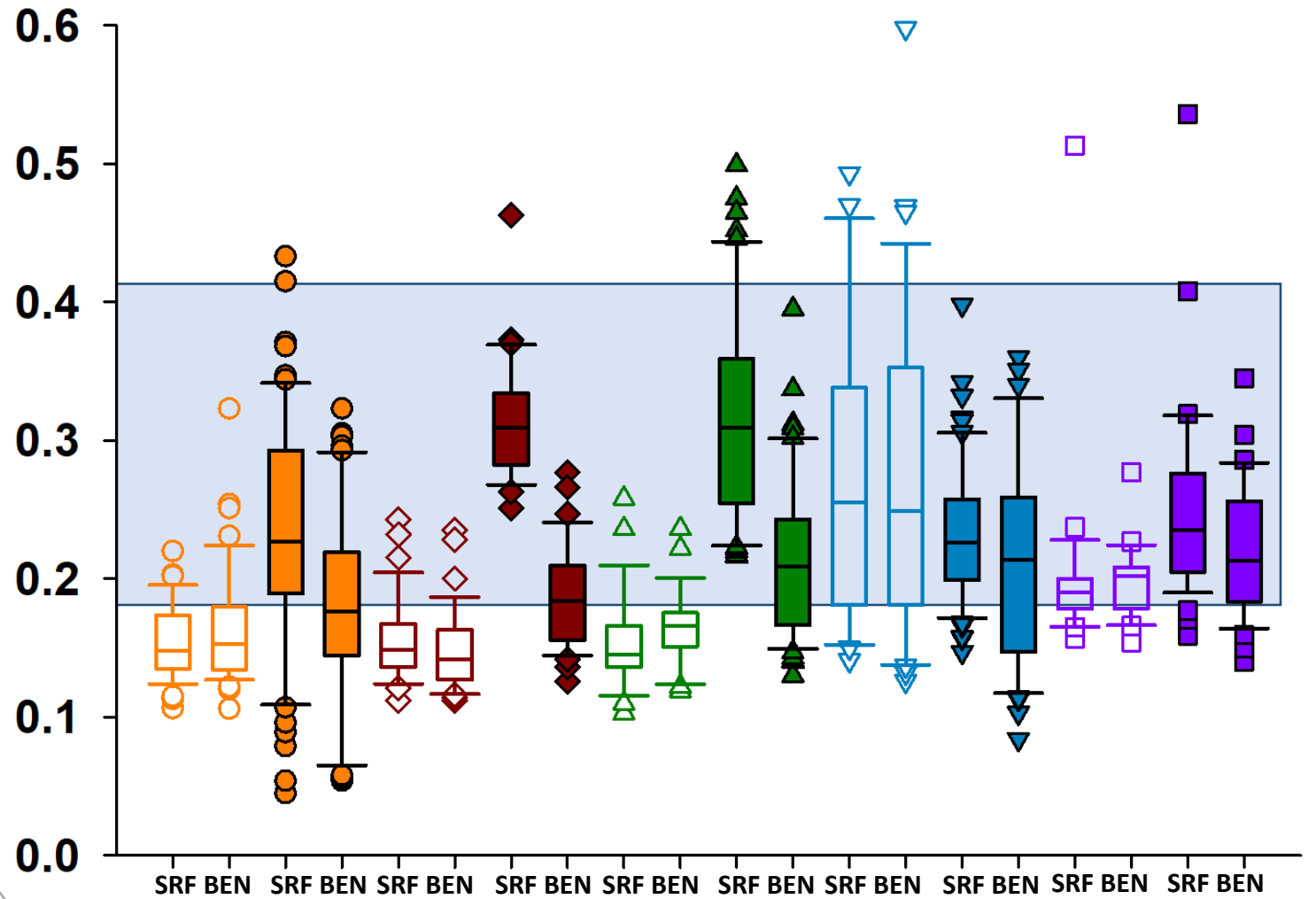


Fall -

Spring -



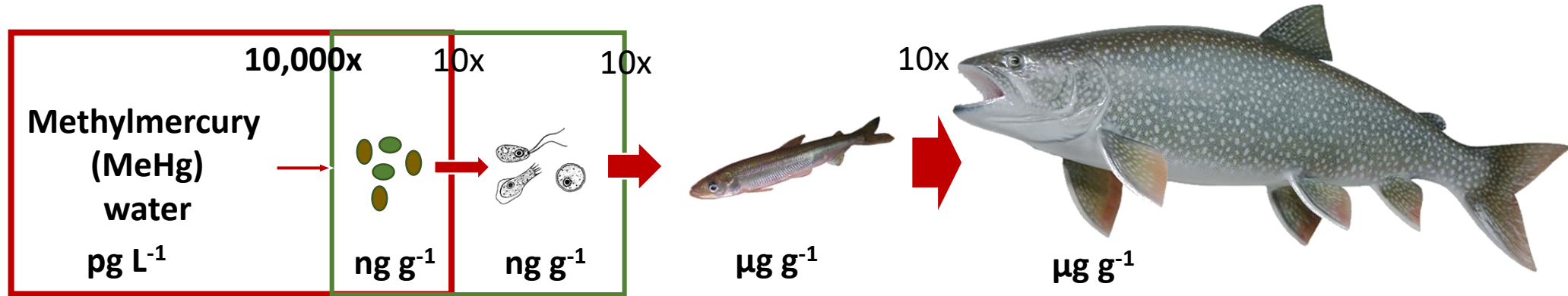
UTHg (ng L<sup>-1</sup>)



- Lack of UTHg buildup in benthos of Upper GLs
  - Sediments are NOT Hg source (Erie maybe)

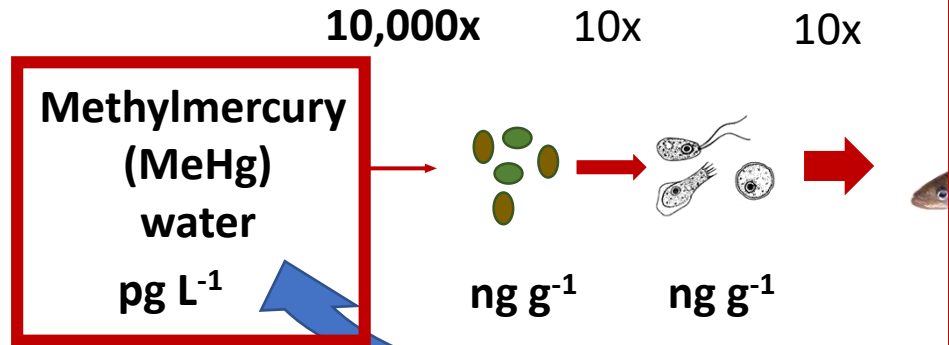


# Hg bioaccumulation and magnification

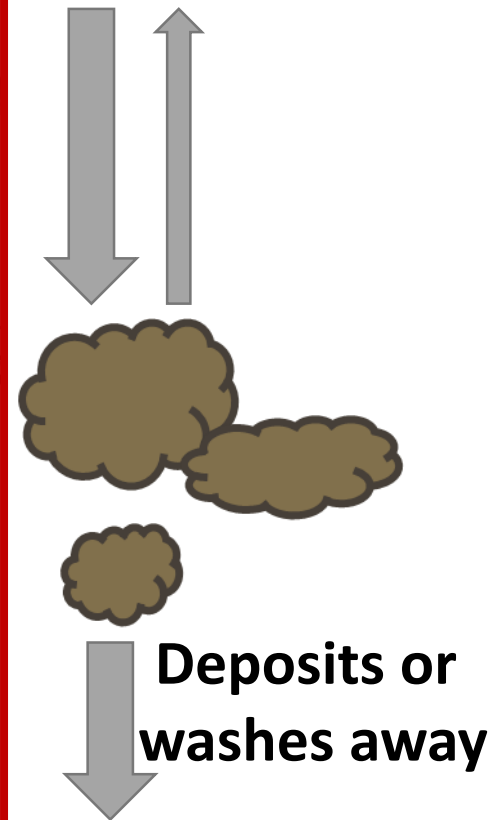


- 1) Are bioaccumulation dynamics enhanced at the critical step?
  - a. Hg concentrations
- 2) Does the lower food web play a substantial role?
  - a. stable isotope ratios - CN

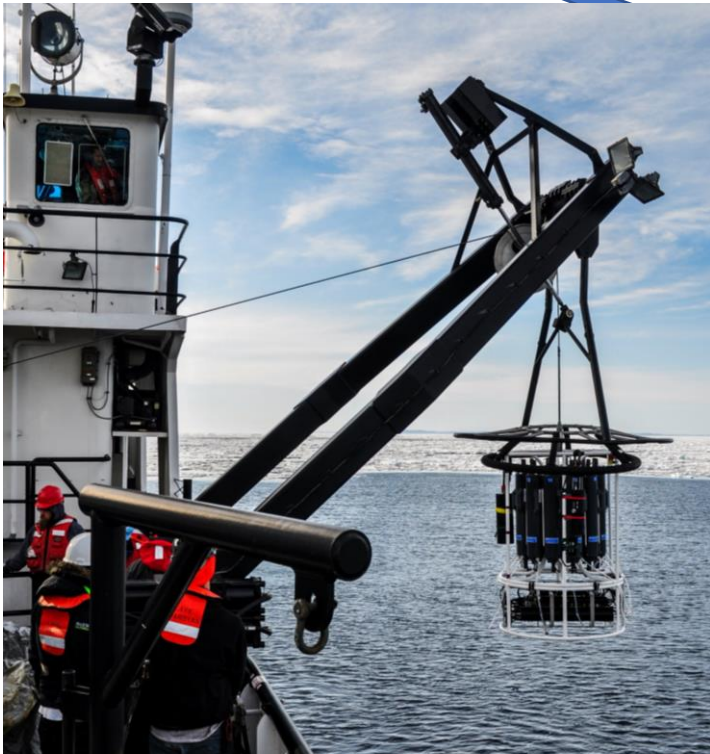
# Hg bioaccumulation and magnification



**Inorganic Hg(II) - Ligand**  
(Delectable)

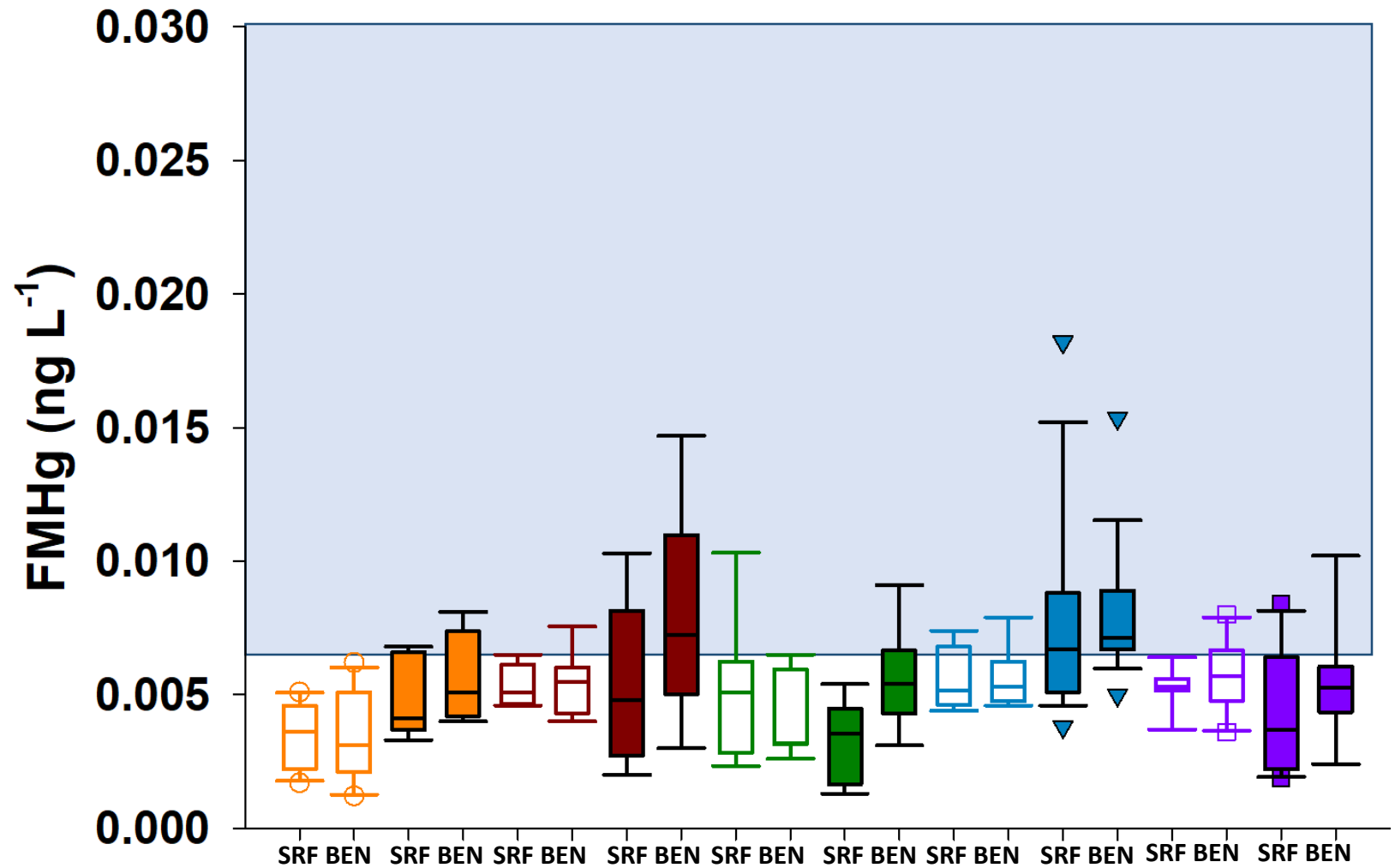
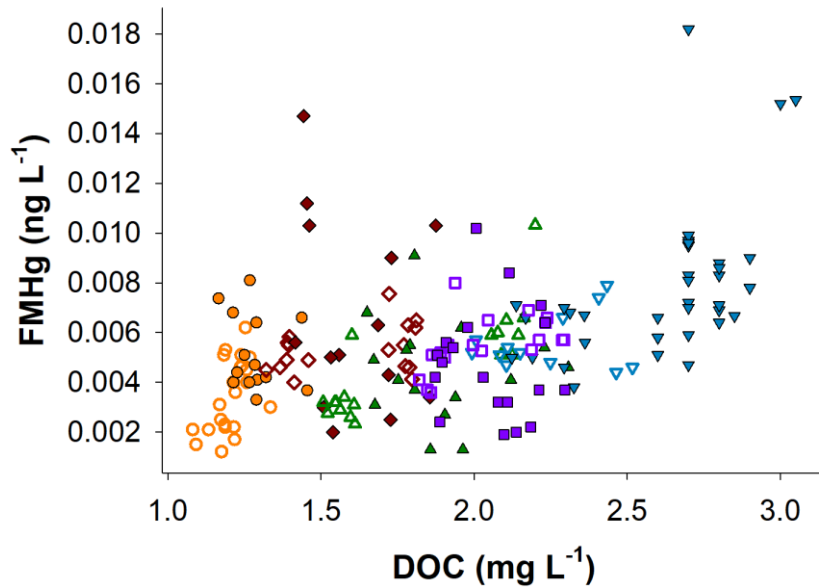


25,000,000 fold high



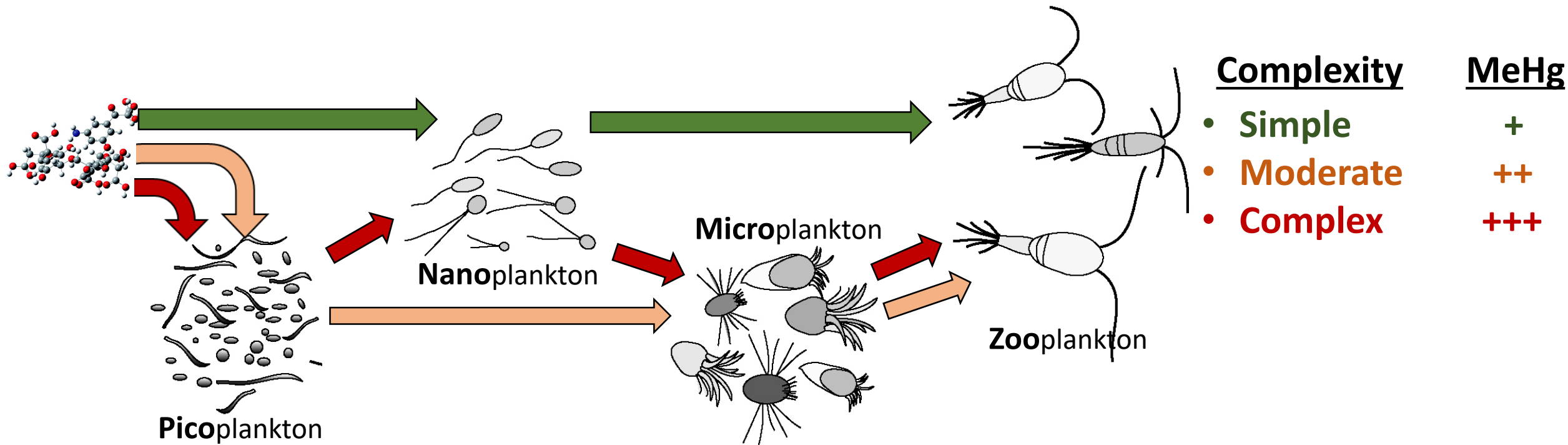
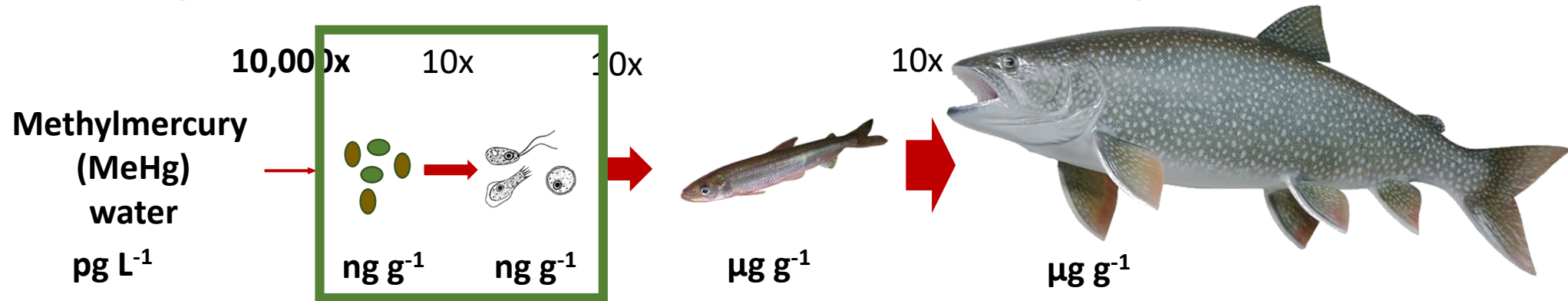
# Dissolved MeHg in water

- 20-30% of MeHg is on particles
- Unlike the ocean, DOC does not modulate FMHg concentrations

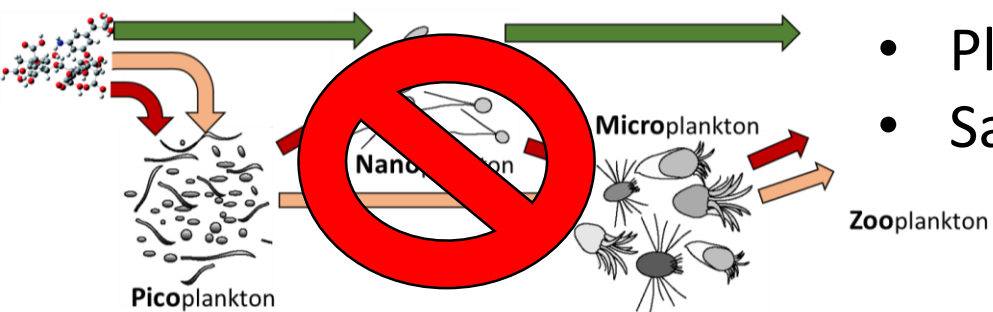
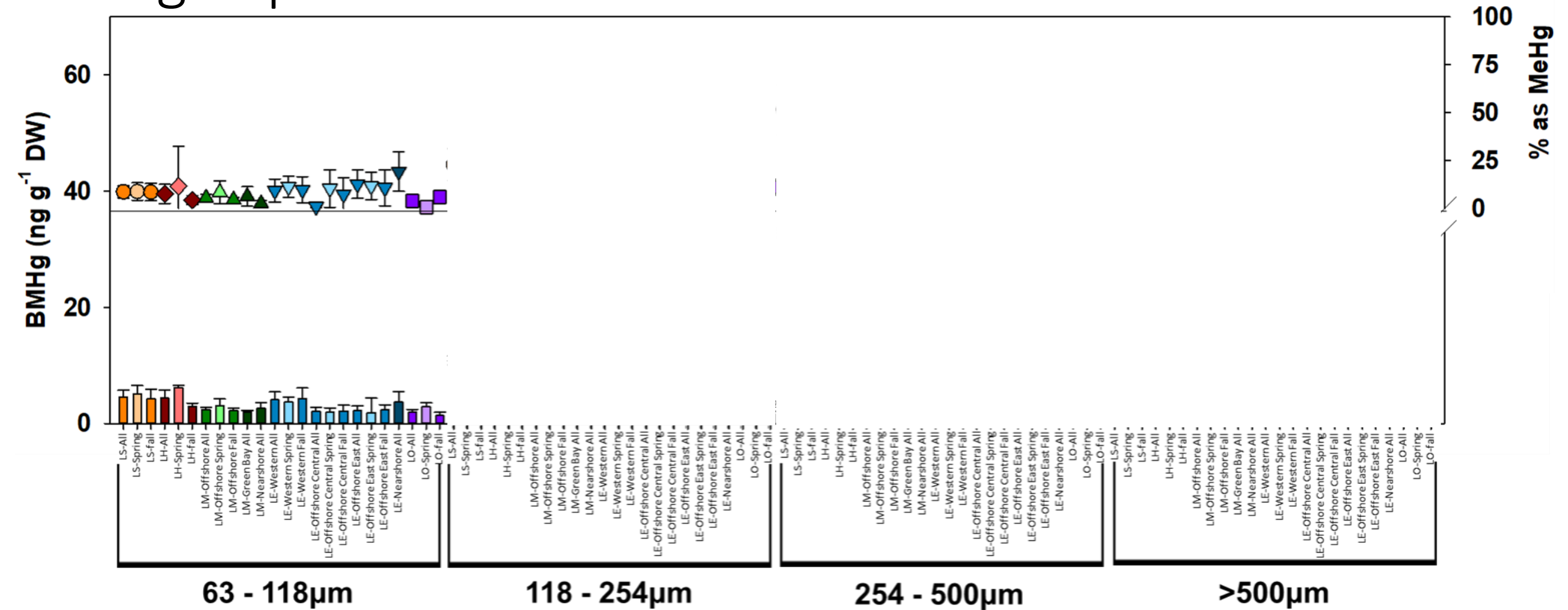


- FMHg buildup in benthos of Upper GLs
  - But 30 – 70% of FMHg is photochemically destroyed in epilimnion

# Hg bioaccumulation and magnification



# MeHg in plankton – food web interactions

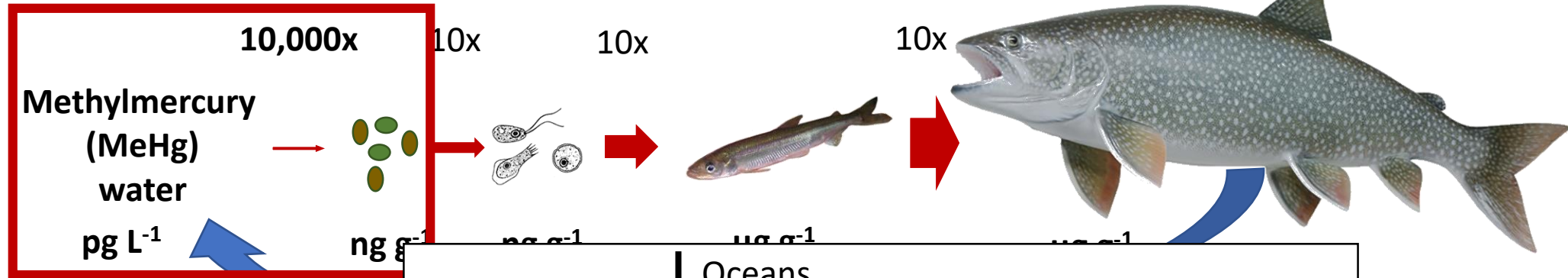


- Planktonic concentrations are 10 – 100 times lower than fish.
- Sampling design captured biomagnification.

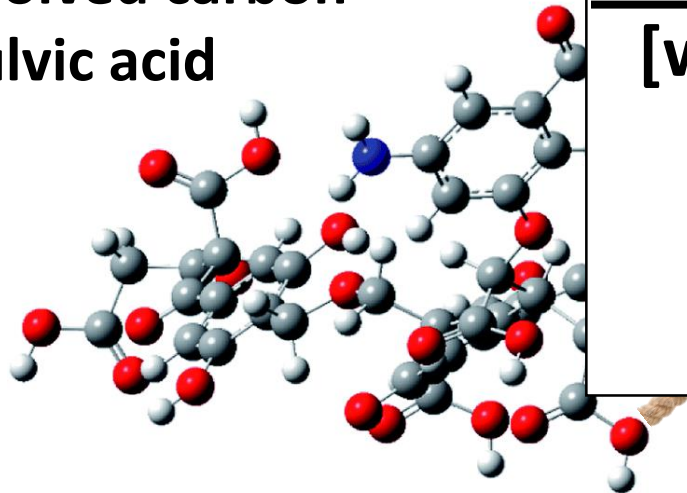
Food web tracers (C and N isotopes) support this.



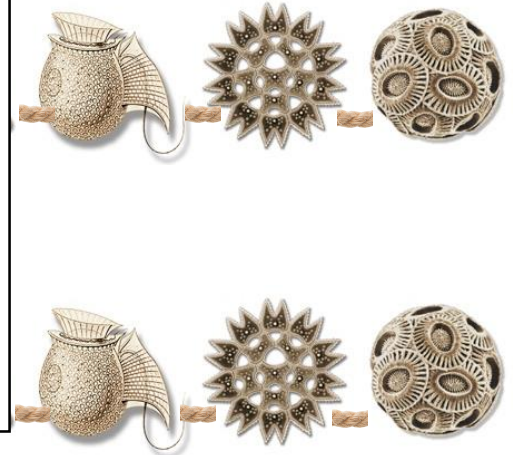
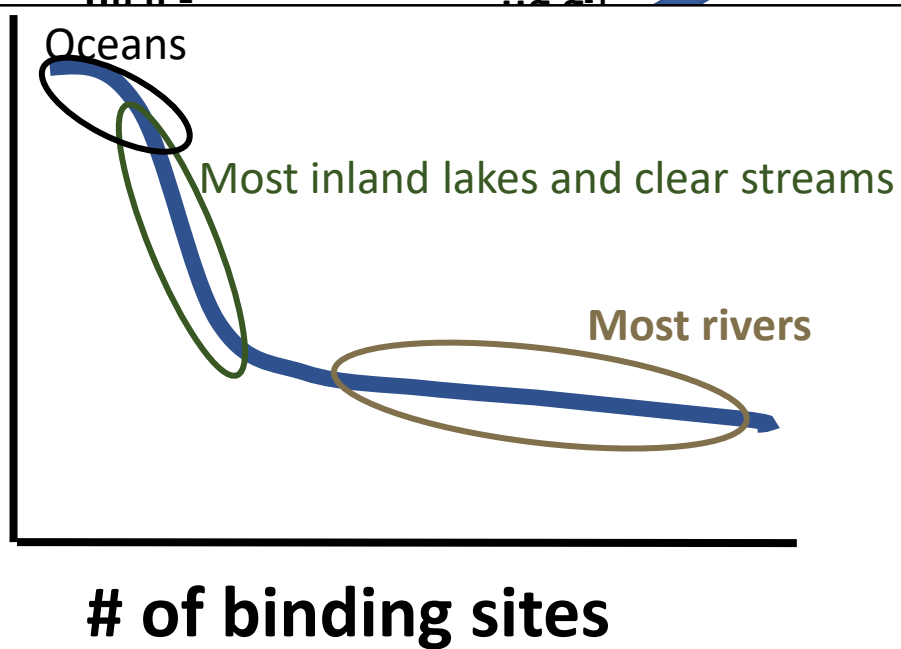
# Hg bioaccumulation and magnification



Dissolved carbon  
– Fulvic acid



**[plankton]**  
**[water]**

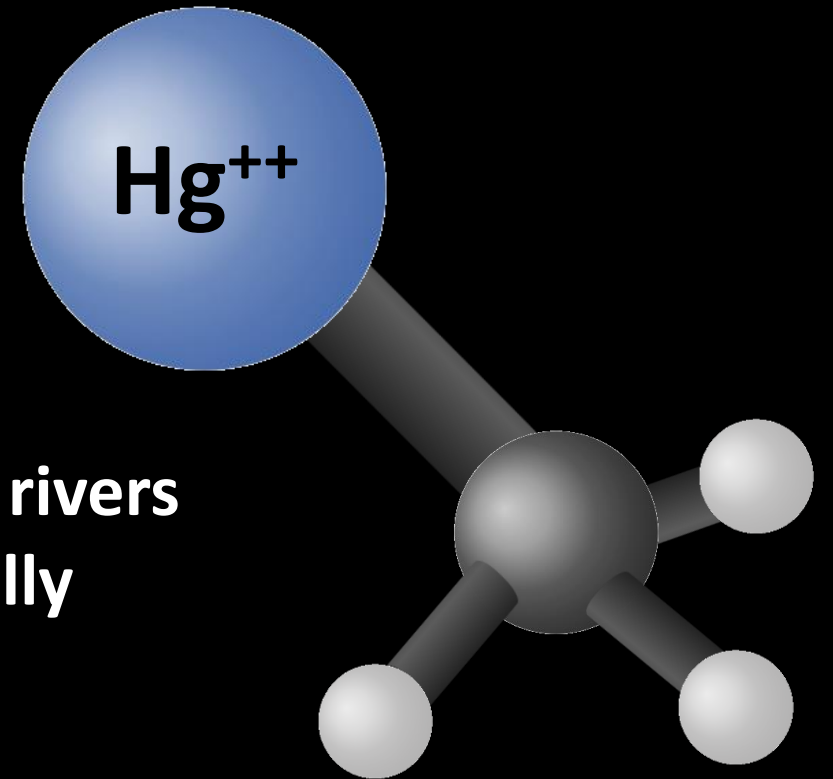


Why ~~Lake Superior is~~ so susceptible to methylmercury accumulation?  
the Great Lakes are

$X^-$  could be:

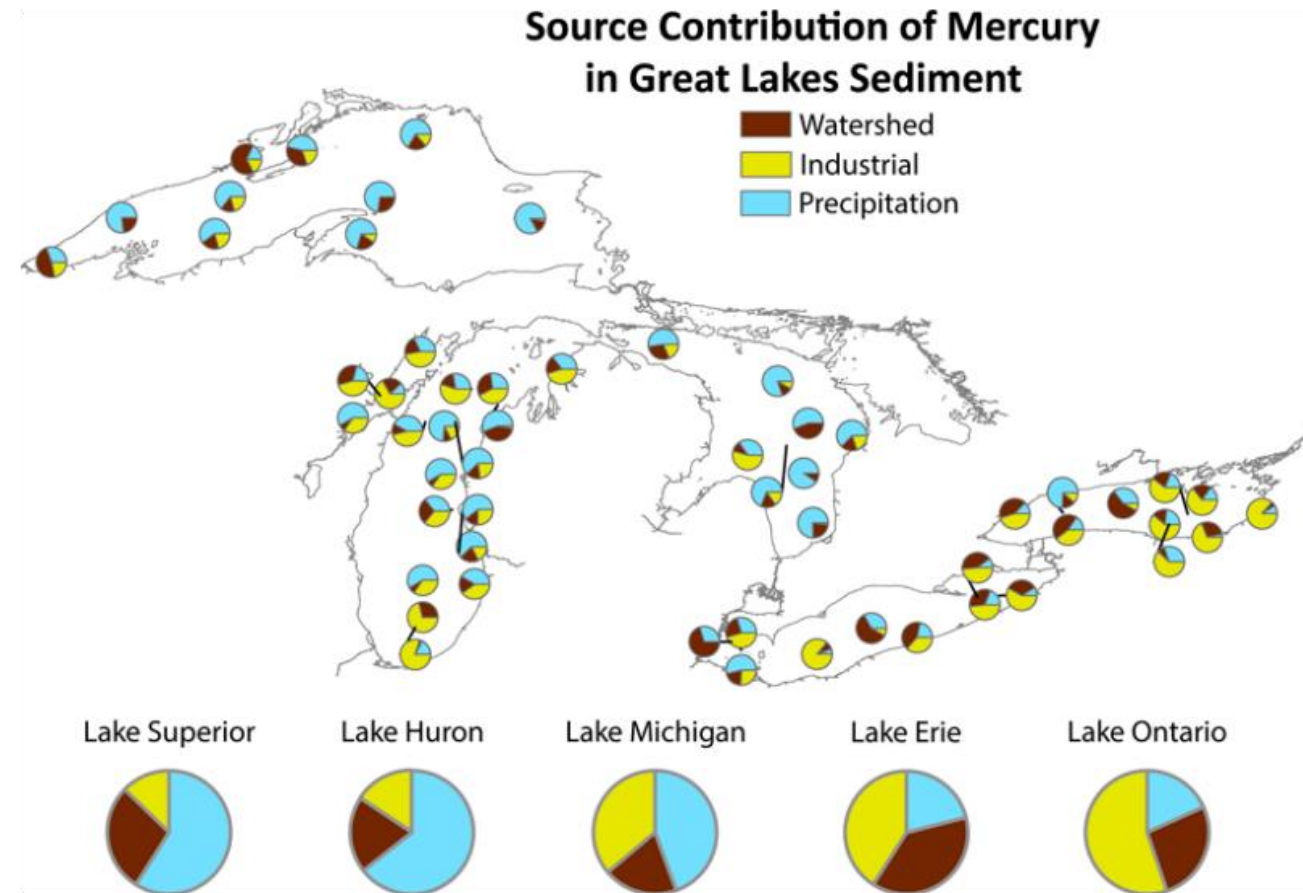
1. Chloride or Bromine – Oceans
2. Large complex DOC molecules – Inland lakes and rivers
3. Small, simple DOC molecules – Great Lakes globally

$X^-$



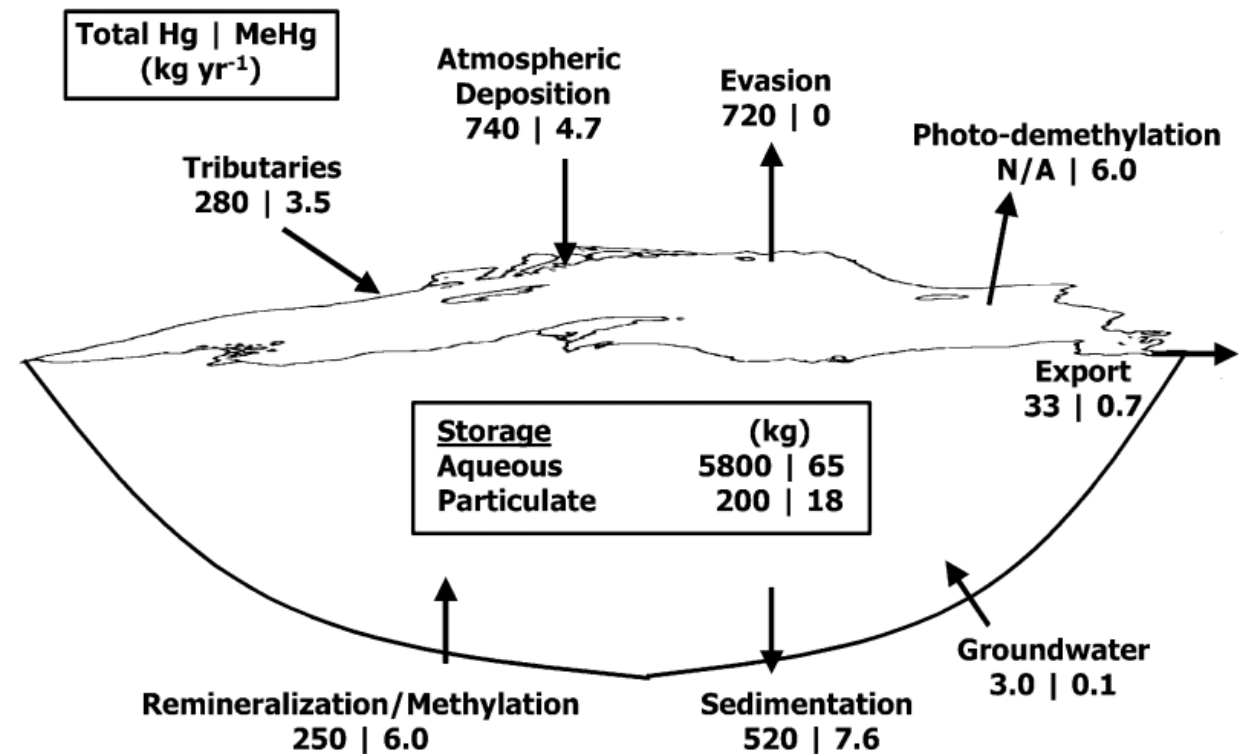
# Great Lakes are susceptible to methylmercury bioaccumulation

- Proportions of atmospherically delivered Hg increase with increased Lake to Watershed ratios and/or minimal littoral zone.
  - GLs span 1.5 – 3.4 (Upper 1.5 – 2.2)<sup>3</sup>
  - Similar to:
    - Baikal - Russia
    - Victoria – E Africa
    - Tanganyika - E Africa
    - Malawi - E Africa
    - Great Bear – Canada
    - Great Slave – Canada
    - Winnipeg – Canada
    - Ladoga – Russia
    - Issyk - Kul – Kyrgyzstan



# Great Lakes are susceptible to methylmercury bioaccumulation

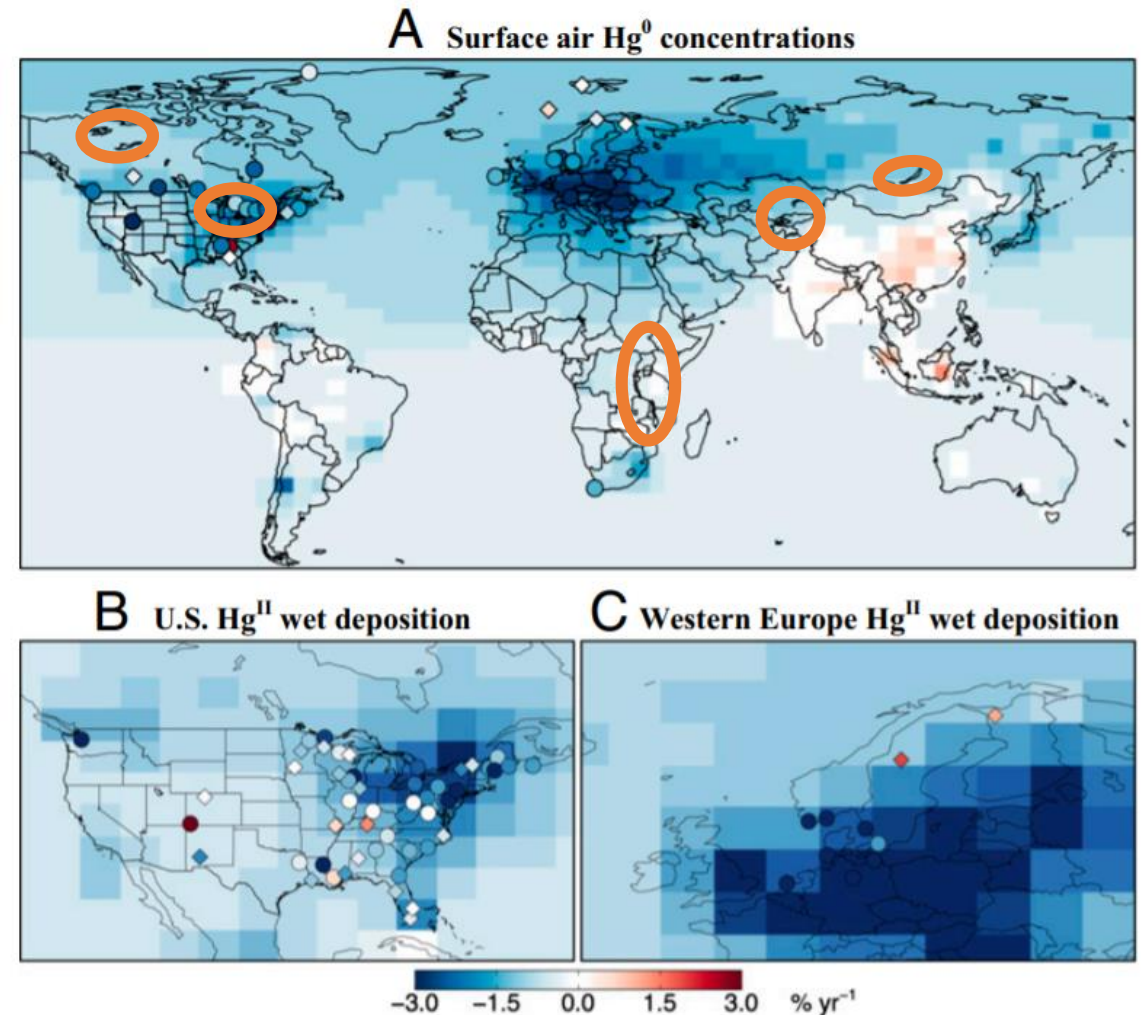
- Proportions of atmospherically delivered Hg increase with increased Lake to Watershed ratios and/or minimal littoral zone.
- These lakes represent the world's most sensitive ecosystems to atmospherically delivered Hg<sup>4</sup>.





# Great Lakes are susceptible to methylmercury bioaccumulation

- Proportions of atmospherically delivered Hg increase with increased Lake to Watershed ratios and/or minimal littoral zone.
- These lakes represent the world's most sensitive ecosystems to atmospherically delivered Hg.
- Thus, they are sentinels to globally declining Hg emissions present in the N. hemisphere<sup>5</sup>.





# Thank you

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*Ryan F. Lepak<sup>a,b</sup>, Jacob M Ogorek<sup>b</sup>, Joel Hoffman<sup>a</sup>, John F DeWild<sup>b</sup>, Michael T. Tate<sup>b</sup>, and David P. Krabbenhoft<sup>b</sup>*

*<sup>a</sup>USEPA GLTED*

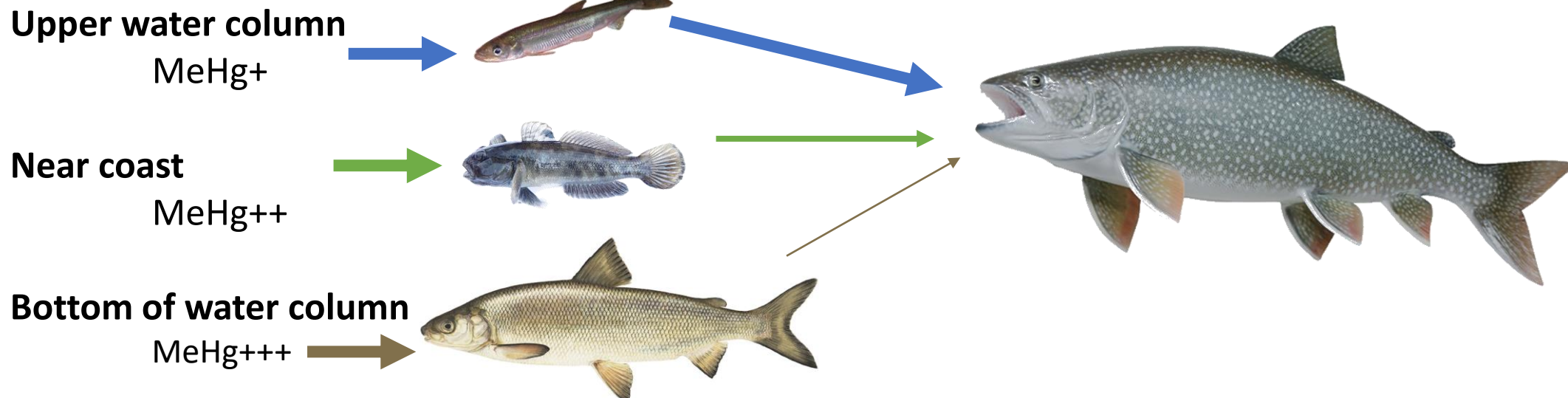
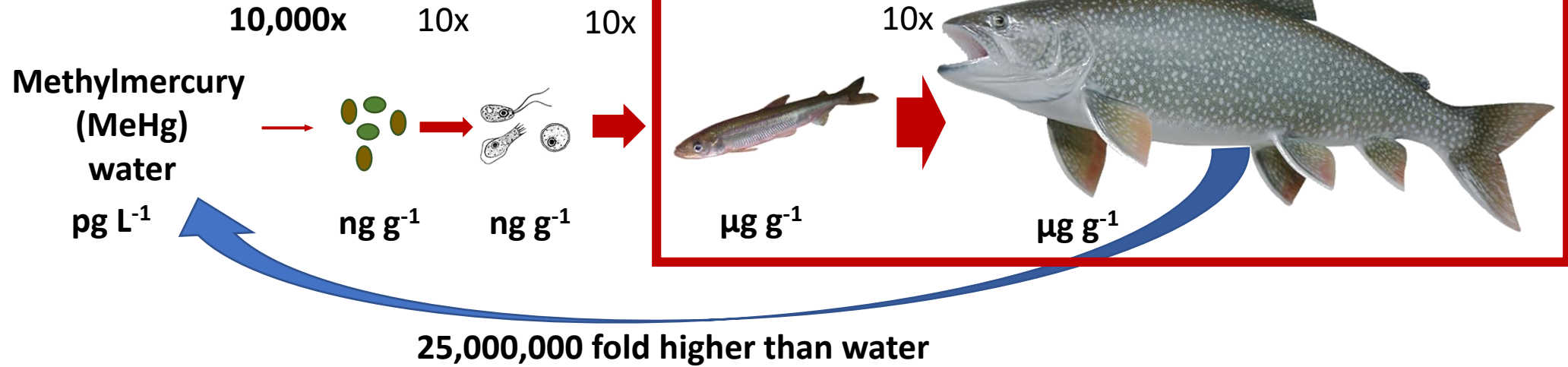
*<sup>b</sup>USGS Hg Research Lab*

# Citations

- 1) Tsui, M. T. K.; Finlay, J. C., Influence of dissolved organic carbon on methylmercury bioavailability across Minnesota stream ecosystems. *Environmental science & technology* **2011**, 45, (14), 5981-5987.
- 2) Schartup, A. T.; Qureshi, A.; Dassuncao, C.; Thackray, C. P.; Harding, G.; Sunderland, E. M., A model for methylmercury uptake and trophic transfer by marine plankton. *Environmental science & technology* **2018**, 52, (2), 654-662.
- 3) Robertson, D. M.; Saad, D. A., Nutrient inputs to the laurentian great lakes by source and watershed estimated using SPARROW watershed models1. *JAWRA Journal of the American Water Resources Association* **2011**, 47, (5), 1011-1033.
- 4) Rolfhus, K.; Sakamoto, H.; Cleckner, L.; Stoor, R.; Babiarz, C.; Back, R.; Manolopoulos, H.; Hurley, J., Distribution and fluxes of total and methylmercury in Lake Superior. *Environmental science & technology* **2003**, 37, (5), 865-872.
- 5) Zhang, Y.; Jacob, D. J.; Horowitz, H. M.; Chen, L.; Amos, H. M.; Krabbenhoft, D. P.; Slemr, F.; Louis, V. L. S.; Sunderland, E. M., Observed decrease in atmospheric mercury explained by global decline in anthropogenic emissions. *Proceedings of the National Academy of Sciences* **2016**, 113, (3), 526-531.



# Hg bioaccumulation and magnification





# Hg bioaccumulation and magnification

