

## Screening the Vast PFAS Landscape: In Vitro Toxicokinetic Testing and LC-MS/MS Analysis

## Marci Smeltz

U.S. Environmental Protection Agency Office of Research and Development Presentation for Waters Southeast Mass Spectrometry Users Meeting

Tuesday, July 21, 2020

The views expressed in this presentation are those of the author and do not necessarily reflect the views or policies of the U.S. EPA



## **EPA Office of Research and Development**

- Office of Research and Development (ORD) is the scientific research arm of the EPA
- Research is conducted by ORD's four Centers
- 10 facilities across the USA
- Research is conducted by a combination of Federal scientists, contract researchers, and postdoctoral, graduate student, and postbaccalaureate trainees



Research Triangle Park, NC



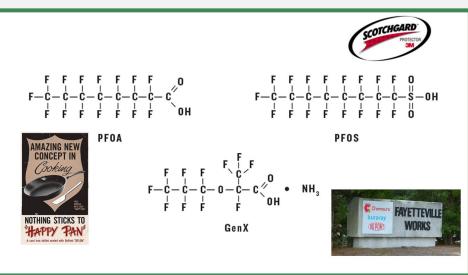


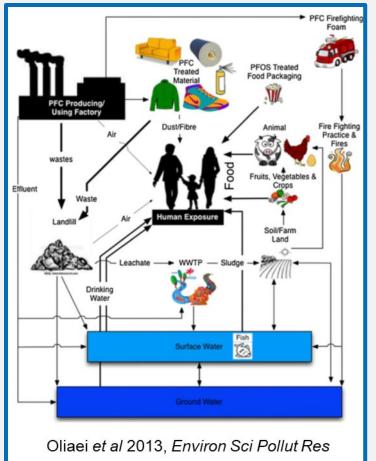
Athens, GA



## What are **PFAS**?

- PFAS = per- and polyfluoroalkyl substances
- <u>Man-made chemicals</u> used in industry and consumer products worldwide since the 1950s
- Repel water, resist heat, and protect surfaces







## Why are PFAS receiving lots of attention?

#### Widespread occurrence

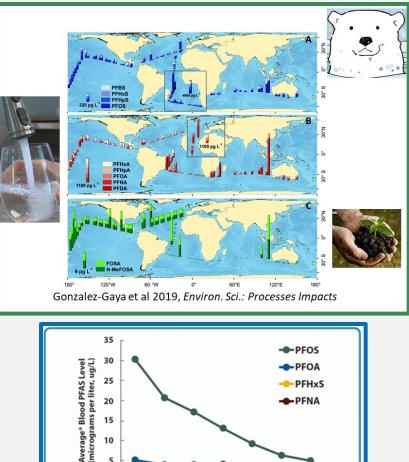
- PFAS in 97% of American population
- Even in arctic polar bears

#### Persistence

- Carbon-fluorine bonds are some of the strongest
- Little degradation in environment

#### Bioaccumulative

- Accumulate over time
- Absorption > elimination



2008 2010

2000

2004

2006

CDC 2017, Fourth Report on Human Exposure

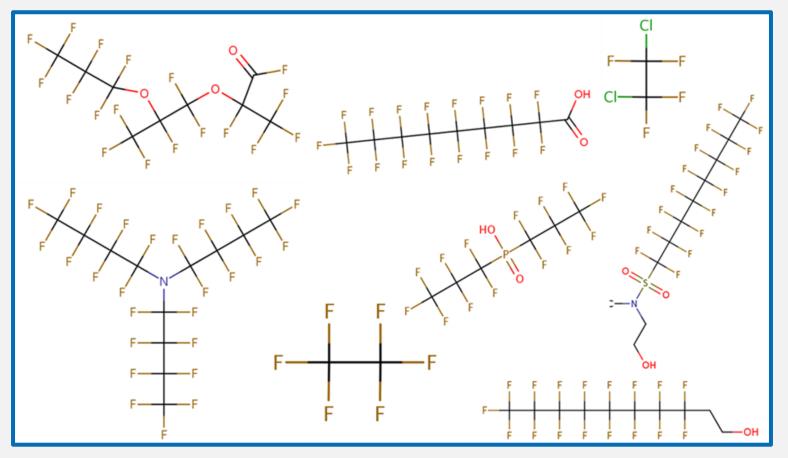
to Environmental Chemicals

2012 2014



## How many PFAS exist?

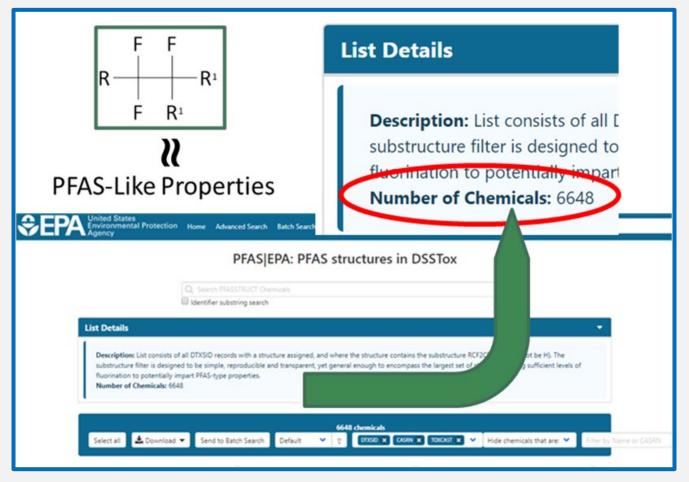
- More than 4700 PFAS recognized by OECD
- As industry continues to invent, the number will increase





## How many PFAS exist?

- More than 4700 PFAS recognized by OECD
- As industry continues to invent, the number will increase

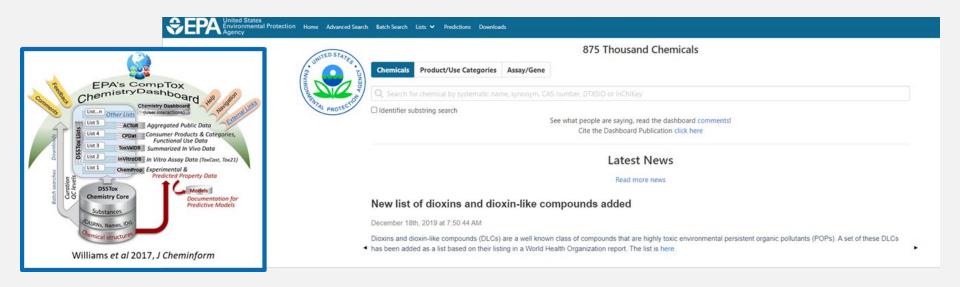


5



## What's the CompTox Chemicals Dashboard?

- https://comptox.epa.gov/dashboard
- One-stop-shop for chemical, toxicological, and exposure information
- Almost 900,000 chemicals inventoried

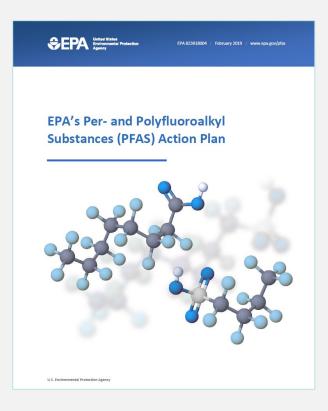




## What is the EPA doing about PFAS?

#### EPA PFAS Action Plan (2019)

- Assist states, tribes, and communities address PFAS with short-term solutions and long-term strategies to address PFAS
- PFAS-Related Challenges
  - Developing/validating laboratory analytical methods for measuring PFAS
  - Assessing PFAS chemical toxicity
    - Developing standard toxicity values for PFAS chemicals
    - Characterizing potential human exposure pathways
    - Managing PFAS containing materials and waste
    - Testing drinking water treatment technologies
    - Identifying site remediation technologies





## Which PFAS are we interested in?

- PFAS Screening Library creation: PFAS Landscape
  - Maximize read-across
  - Capture structural diversity
- 1220 PFAS currently in TSCA inventory

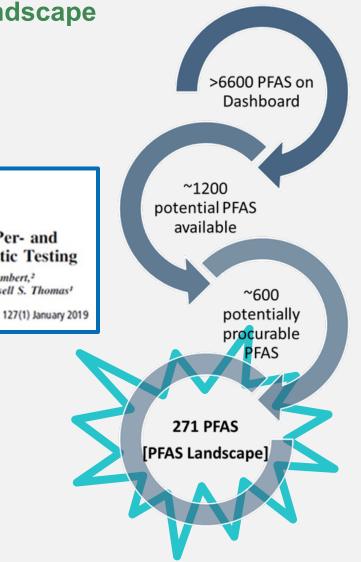
**Brief Communication** 

A Chemical Category-Based Prioritization Approach for Selecting 75 Per- and Polyfluoroalkyl Substances (PFAS) for Tiered Toxicity and Toxicokinetic Testing

Grace Patlewicz,<sup>1</sup> Ann M. Richard,<sup>1</sup> Antony J. Williams,<sup>1</sup> Christopher M. Grulke,<sup>1</sup> Reeder Sams,<sup>1</sup> Jason Lambert,<sup>2</sup> Pamela D. Noyes,<sup>3</sup> Michael J. DeVito,<sup>4</sup> Ronald N. Hines,<sup>5</sup> Mark Strynar,<sup>6</sup> Annette Guiseppi-Elie,<sup>6</sup> and Russell S. Thomas<sup>1</sup>

**Environmental Health Perspectives** 

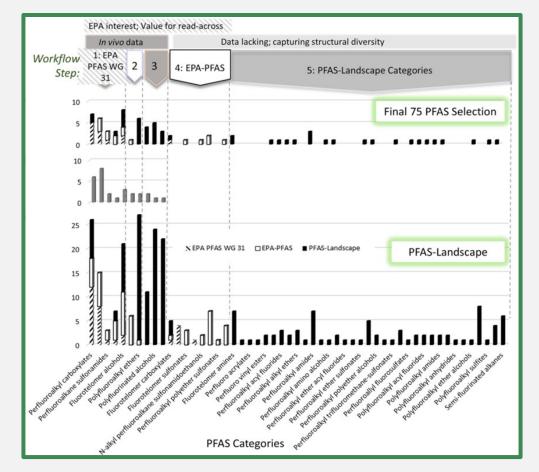
014501-1





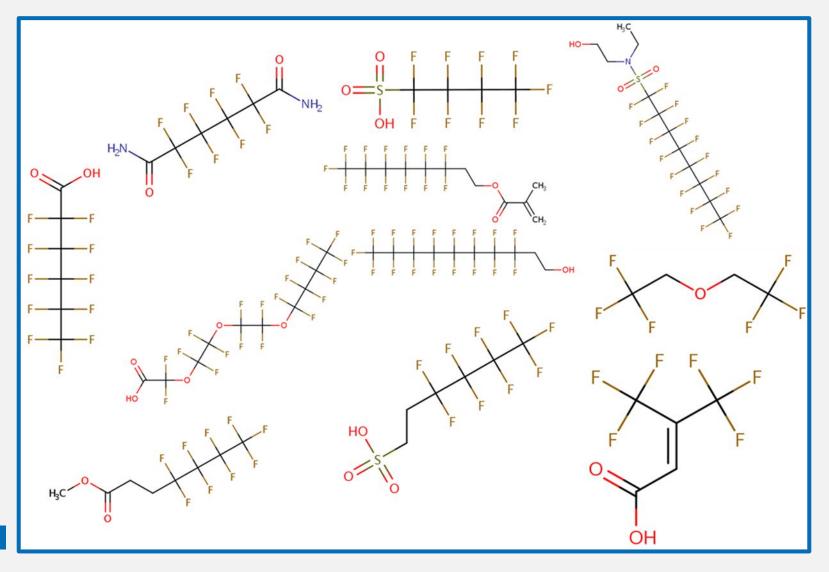
## Which PFAS are we interested in?

Initially, 75 PFAS selected from the PFAS Landscape, but the PFAS of interest now expands to near <u>200</u> unique structures and the Landscape up to 430 unique PFAS





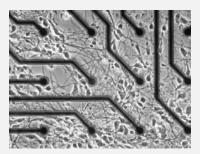
### Which PFAS are being evaluated?

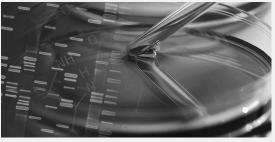




## How are we examining these PFAS?

- A range of targeted and tiered high-throughput toxicity assays to serve as guide for potential human health risk
- New approach methodologies (NAMs) used
  - Alternative test methods and strategies to reduce, refine, and/or replace mammalian animals
  - In vitro tests/assays, in chemico assays, in silico algorithms
- Endpoints for PFAS work
  - Hepatotoxicity
  - Immunotoxicity
  - Developmental toxicity
  - Mitochondrial toxicity
  - In vitro toxicokinetic assays

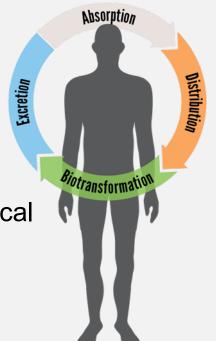






## What are in vitro toxicokinetic assays?

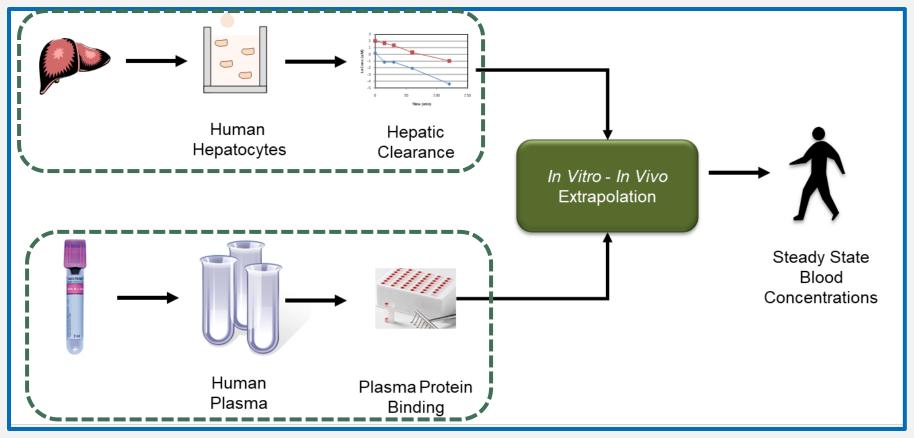
- Toxicokinetics: the study of how a substance gets into the body and what happens to it in the body
  - Can be used to look at how chemicals move throughout the body and lead to harmful effects
  - Often viewed as a function of dose over time
- Kinetic data can inform...
  - Bioavailability (degree of a substance to enter circulation when introduced to body)
  - Bioaccumulation potential (absorption >> excretion)
  - Metabolite formation (transformation of original chemical new entity; can lead to bioactivation or detoxification)





## What are in vitro toxicokinetic assays?

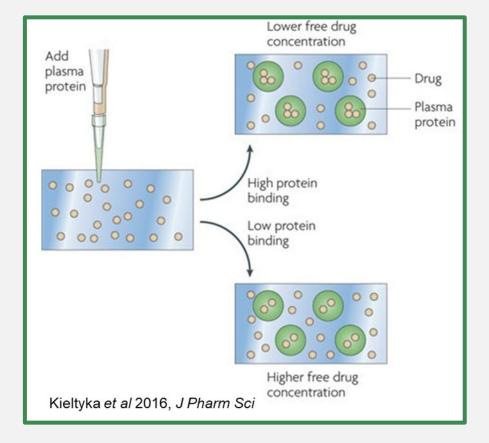
 Toxicokinetics: the study of how a substance gets into the body and what happens to it in the body





## What is plasma protein binding?

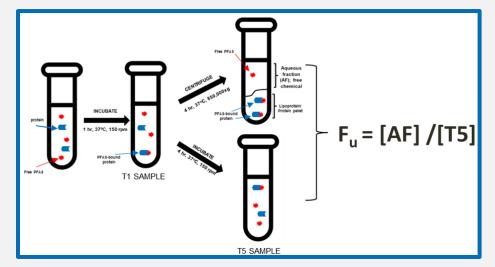
- Assay to assess the free (unbound) fraction of chemical to proteins within the blood
  - F<sub>u</sub>
  - Unbound molecules permeate through cell membranes to reach 'target'
  - Determine by equilibrium dialysis, ultrafiltration, and/or ultracentrifugation
- Ultracentrifugation assay used
  - Human plasma (10-donor pool, mixed sex) centrifuged to separate aqueous fraction from albumin, lipoproteins, and fatty acids
  - Mixtures of up to 4 PFAS (10 µM) were included with each plasma sample, run in triplicate





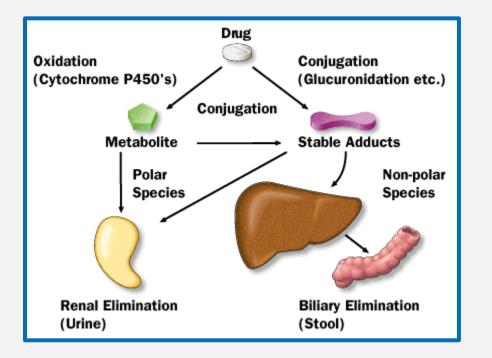
## What is plasma protein binding?

- Assay to assess the free (unbound) fraction of chemical to proteins within the blood
  - F<sub>u</sub>
  - Unbound molecules permeate through cell membranes to reach 'target'
  - Determine by equilibrium dialysis, ultrafiltration, and/or ultracentrifugation
- Ultracentrifugation assay used
  - Human plasma (10-donor pool, mixed sex) centrifuged to separate aqueous fraction from albumin, lipoproteins, and fatty acids
  - Mixtures of up to 4 PFAS (10 µM) were included with each plasma sample, run in triplicate





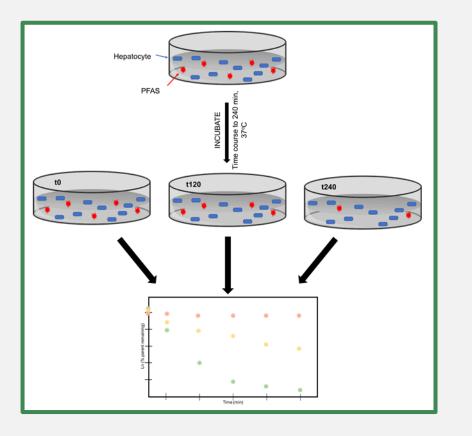
## What is in vitro hepatic clearance?



- Liver is the major site of drug metabolism in body (hepatic)
- Hepatic clearance (CL<sub>hepatic</sub>) is measure of the rate of elimination of a chemical from the liver
- Models to study metabolism:
  - Human liver microsomes
  - Recombinantly expressed enzymes
  - Hepatocytes contain full complement of hepatic drug metabolizing enzymes



### What is *in vitro* hepatic clearance?



For the PFAS work...

- Substrate depletion approach using primary human hepatocytes (50-donor pool, mixed sex) at 1 µM PFAS concentration
- Time course: 0, 15, 30, 60, 90, 120, and 240 min with non-linear regression fit
- Work completed by collaborator at National Toxicology Program (NIEHS) [David Crizer]



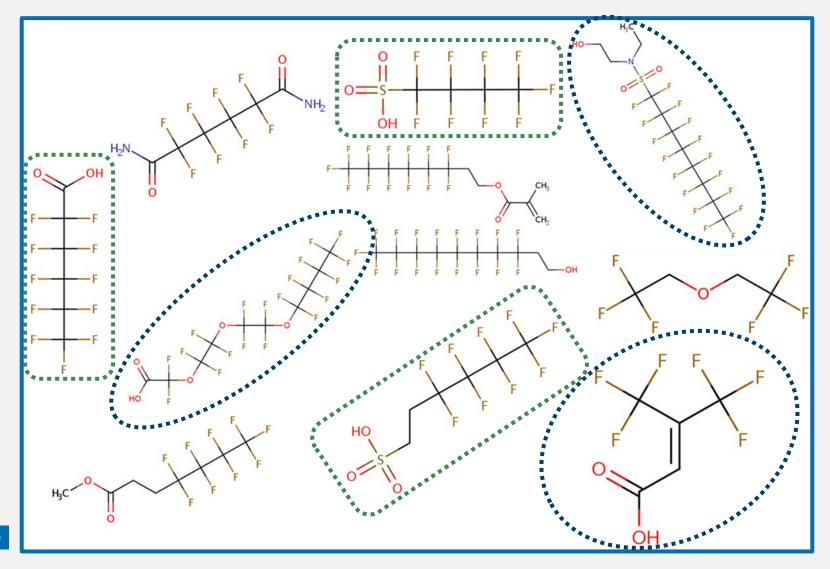
## How do we analyze these assay samples?

- Both assays require concentration determination of parent PFAS
- EPA has a range of analytical capabilities (single quads, triple quads, high resolution mass spec)





### Which PFAS of interest are LC-able?





# What guided our LC-MS/MS method development journey?

CEPA United States Approximated Protection Approximated Protection	[APPLICATION NOTE]	Waters THE SCIENCE OF WHAT'S POSSIBLE"				
METHOD 533: DETERMINATION OF PER- AND POLYFLUOROALKYL SUBSTANCES IN DRINKING WATER BY ISOTOPE DILUTION ANION EXCHANGE SOLID PHASE EXTRACTION AND LIQUID CHROMATOGRAPHY/TANDEM		echnique for Perfluorinated Alkyl Substance g UniSpray for Water and Soil Samples				
MASS SPECTROMETRY		[TECHNOLOGY BRIEF] Waters				
METHOD 537.1 DETERMINATION OF SELE POLYFLUORINATED ALKY WATER BY SOLID PHASE E CHROMATOGRAPHY/TAND (LC/MS/MS)	L SUBSTANCES IN DRINKING KTRACTION AND LIQUID	Ultra Low-Level Detection of Perfluoroalkyl Substances (PFASs) Using the PFC Analysis Kit Lauren Mullin and Jennifer Burgess Analytical and Bioanalytical Chemistry (2019) 411:3507-3520 https://doi.org/10.1007/s00216-019-01829-8 RESEARCH PAPER				
ISO 21675:2019 Water quality — Determination of perfluoroa polyfluoroalkyl substances (PFAS) in water – using solid phase extraction and liquid chron tandem mass spectrometry (LC-MS/MS)	- Method	A single analytical method for the determination of 53 legacy and emerging per- and polyfluoroalkyl substances (PFAS) in aqueous matrices Timothy L. Coggan <sup>1</sup> • Tarun Anumol <sup>2</sup> · James Pyke <sup>2</sup> · Jeff Shimeta <sup>1</sup> · Bradley O. Clarke <sup>1</sup>				
20		Identification of Per- and Polyfluoroalkyl Substances in the Cape Fear River by High Resolution Mass Spectrometry and Nontargeted Screening				

James McCord<sup>†</sup><sup>©</sup> and Mark Strynar<sup>\*,‡</sup><sup>©</sup>



21

## What LC-MS/MS settings were used?

#### **ACQUITY UPLC I-Class FTN**

- Equipped with Waters PFC Kit
- CORTECS T3 2.7 µM 3.0x100 mm
- Column Temp: 55°C
- Flow Rate: 0.6 mL/min
- Run Time: 6.5 min
- Mobile Phase A: 95:5 water: acetonitrile with 2.5 mM ammonium acetate
- Mobile Phase B: 5:95 water: acetonitrile with 2.5 mM ammonium acetate



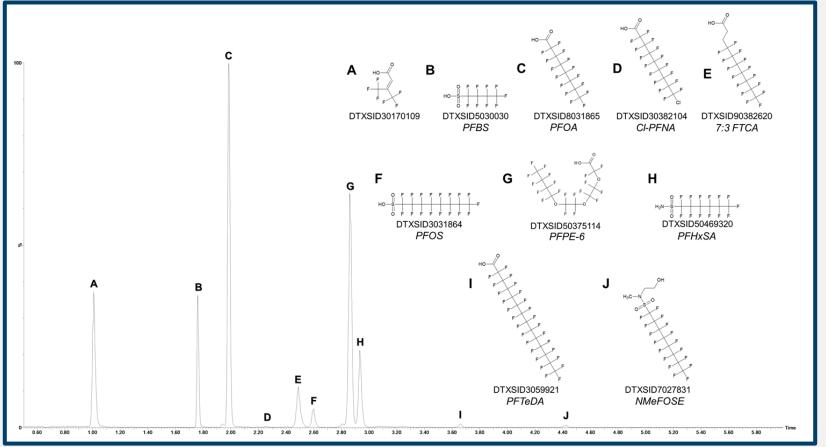
#### Xevo TQ-S Micro

- MRM transitions determined
- Acquisition Polarity: ESI+ and ESI-
- Capillary Voltage: 0.4 kV
- Source Temperature: 150°C
- Desolvation Temperature: 500°C
- Desolvation Gas Flow: 1000 L/hr
- Cone Gas Flow: 150 L/hr
- 19 mass-labelled PFAS (Wellington Laboratories, MPFAC-24ES) was included for quantitation





## How's the method working?

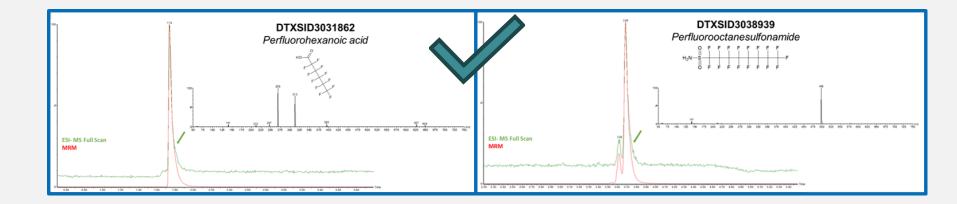


- Mixture of PFAS run at 100 ppb
- Most have estimated LOQ < 50 ppt</p>
- Most PFAS were analyzed in ESI negative; others were monitored as acetate adducts, fragmented in-source



## What about DMSO stock quality?

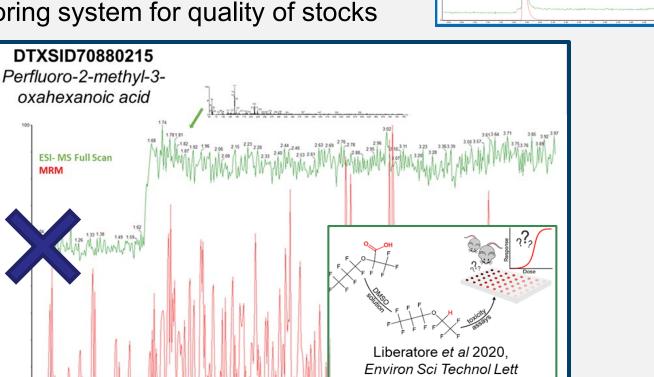
- DMSO is a common solvent used for in vitro assays
- RADAR = MRM (MS/MS) + MS full scan
  - Monitor for any interferences and impurities
  - Application Note: 720005033EN
- Created scoring system for quality of stocks





## What about DMSO stock quality?

- DMSO is a common solvent used for *in vitro* assays
- RADAR = MRM (MS/MS) + MS full scan
  - Monitor for any interferences and impurities
  - Application Note: 720005033EN
- Created scoring system for quality of stocks



2.60

FSL MS Full Sca

DTXSID3031862 Perfluorohexanoic acid

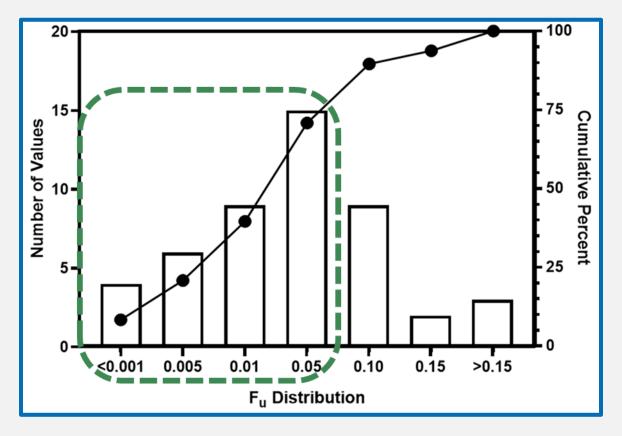
24



## What did we find from the plasma protein binding assay?

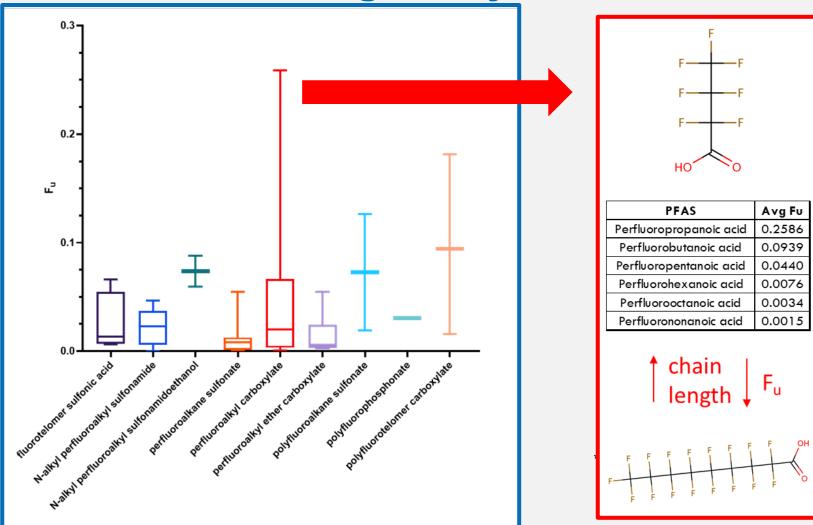
#### 50 LC-able PFAS have determined fraction unbound data

 $\mathbf{F}_{\mathbf{u}} \downarrow \mathbf{binding} \text{ to plasma proteins} \uparrow$ 



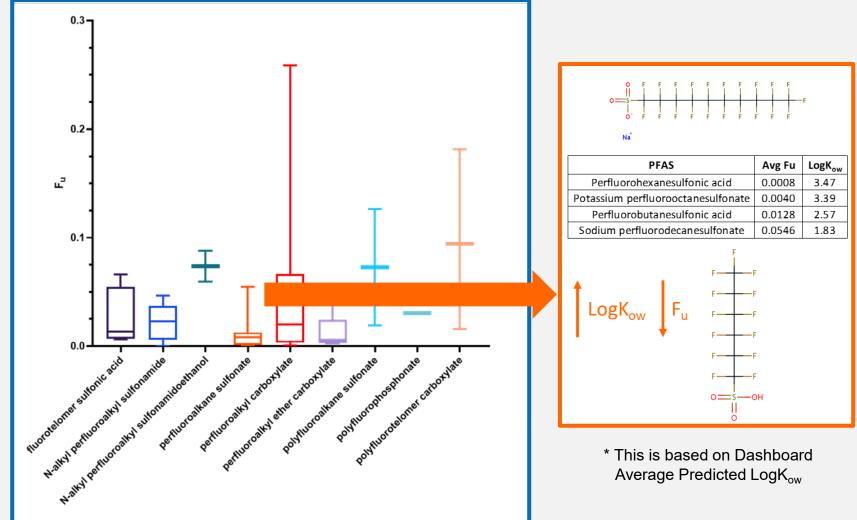


## What did we find from the plasma protein binding assay?





# What did we find from the plasma protein binding assay?

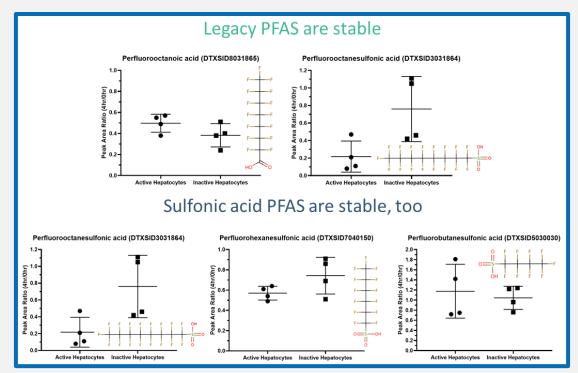


27



# Any observations from the hepatic clearance assay?

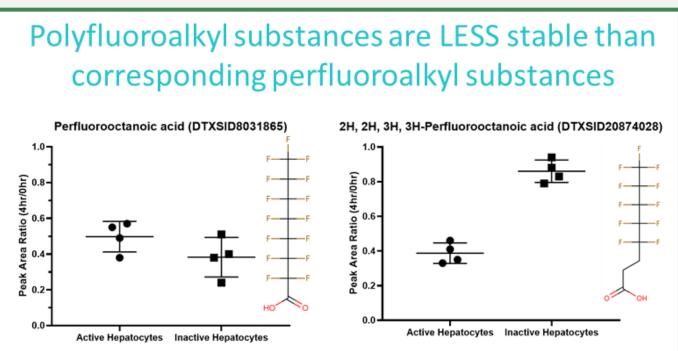
- More than 20 LC-able PFAS assessed
- 1. In vitro hepatic clearance screen
  - 0 and 4 hr time points for active and inactive hepatocytes
  - Compared time ratios to examine for clearance potential





# Any observations from the hepatic clearance assay?

- More than 20 LC-able PFAS assessed
- 1. In vitro hepatic clearance screen
  - 0 and 4 hr time points for active and inactive hepatocytes
  - Compared time ratios to examine for clearance potential

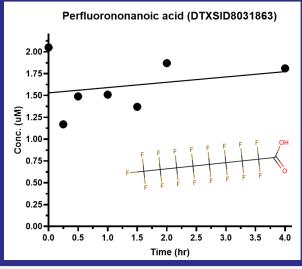


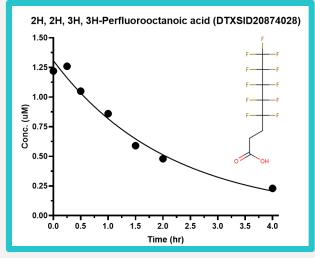


# Any observations from the hepatic clearance assay?

- More than 20 LC-able PFAS assessed
- 2. Metabolic stability time course
  - 0, 0.25, 0.50, 1, 1.5, 2, 4 hr time points
  - Non-linear fit to determine half-life (T<sub>1/2</sub>)

-	Compound Name	Half-life (min)	Clearance (µL/min/million cells)
	Perfluorobutanoic acid	44769343	1.55E-05
	Potassium perfluorohexanesulfonate	21340366	3.25E-05
	Perfluorohexanoic acid	237257	2.92E-03
	Ammonium perfluorooctanoate	88735	7.81E-03
	Potassium perfluorobutanesulfonate	2300	3.01E-01
	Perfluorononanoic acid	1155	6.00E-01
	Perfluorooctanesulfonic acid	990	7.00E-01
	Perfluoro(4-methoxvbutanoic) acid	346.5	2.00E+00
	2H,2H,3H,3H-Perfluorooctanoic acid	101.4	6.83E+00
	N-Ethylperfluorooctanesulfonamide	57	1.22E+01
	3-(Perfluoro-2-butyl)propane-1,2-diol	35.87	1.93E+01
	Perfluoro-3,6,9-trioxatridecanoic acid	29.71	2.33E+01
	Nonafluoropentanamide	25.45	2.72E+01
	3,3-Bis(trifluoromethyl)-2-propenoic acid	19.77	3.51E+01
	4:2 Fluorotelomer sulfonic acid	17.5	3.96E+01
	Octafluoroadipamide	12.8	5.41E+01
	Perfluoropentanamide	10.63	6.52E+01
	N-Methylperfluorooctanesulfonamide	10.17	6.81E+01
	2,2,3,3,4,4-Hexafluorobutanoic acid	4.209	1.65E+02
	Perfluorooctanesulfonamide	2.789	2.48E+02





Clearance rate increasing (faster)

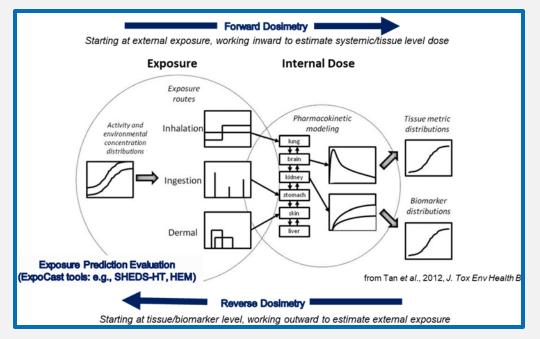


### What is IVIVE?

- In vitro-in vivo extrapolation = IVIVE
  - Model approach that allows *in vitro* data to be extrapolated to estimate corresponding *in vivo* effects
  - Start at tissue/biomarker level  $\rightarrow$  estimate external exposure

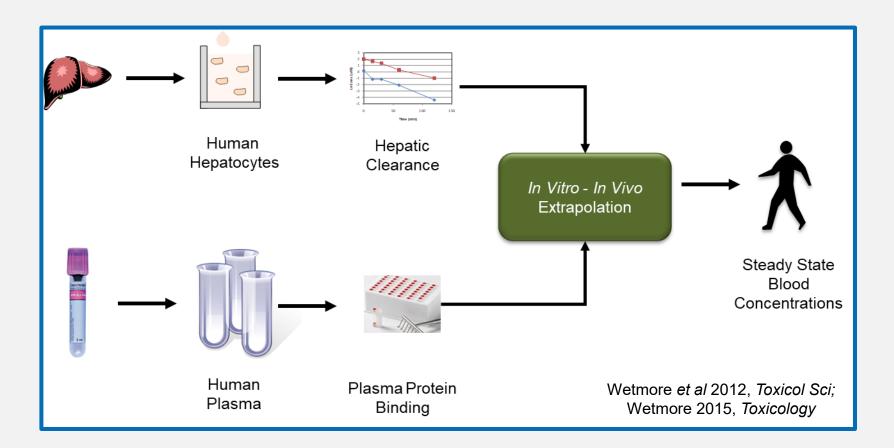
#### Steady-state concentration (C<sub>ss</sub>)

- Concentration of compound in body that stays consistent
- This takes into account plasma protein binding and hepatic clearance data



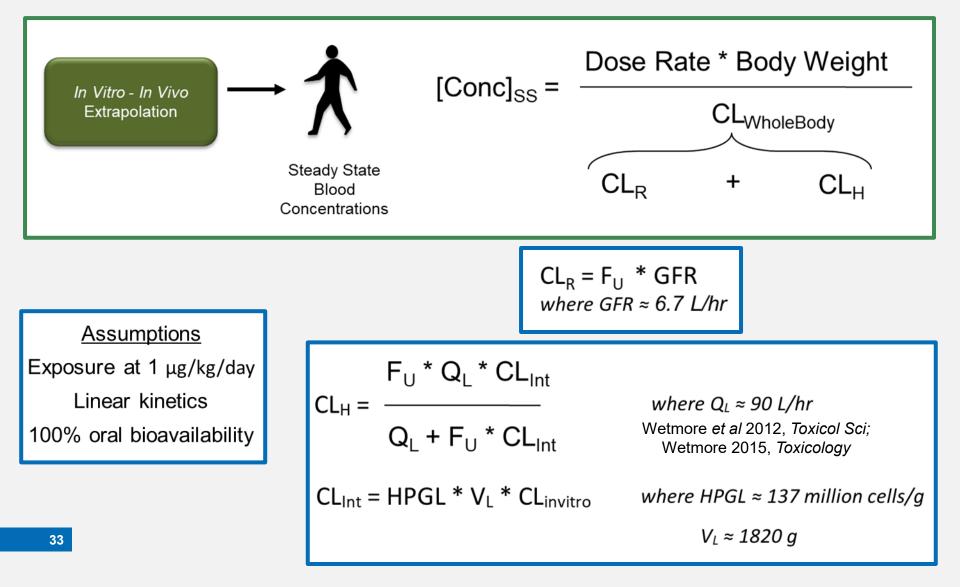


### What is IVIVE?



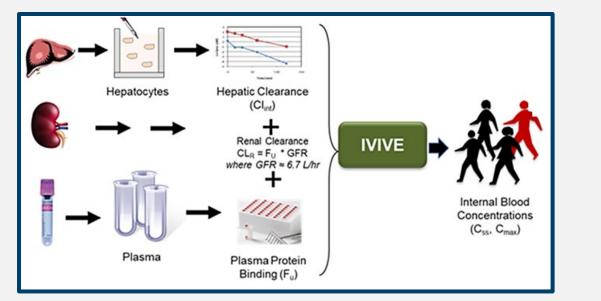


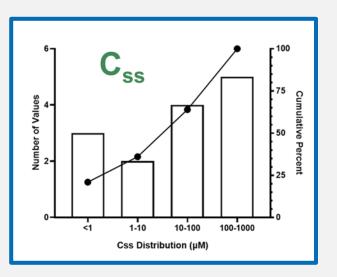
### What is IVIVE?





### What did IVIVE show with PFAS data?





NH

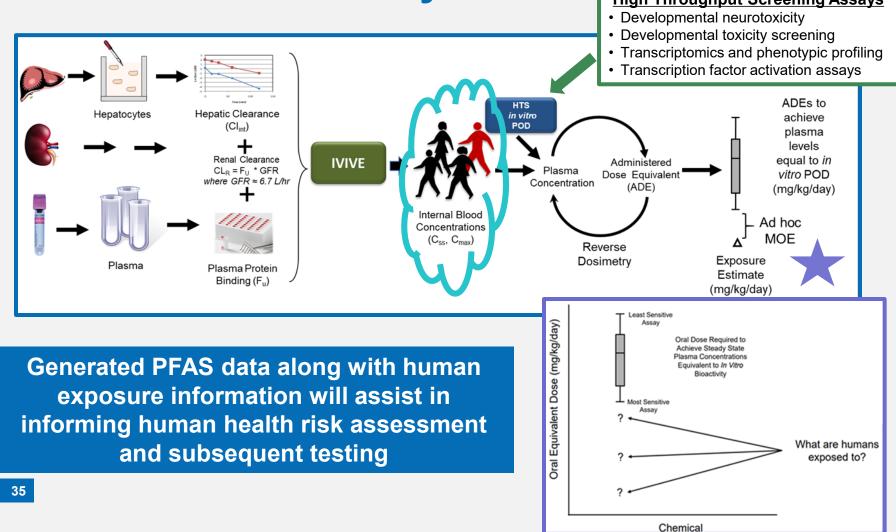
0

Compound Name	Fu	Cl <sub>renal</sub> (L/hr)	Cl <sub>hepatic</sub> (L/hr)	Css (µM)		
assium perfluorohexanesulfonate	0.0011	0.0075	3.82E-07	894.5132		
Ammonium perfluorooctanoate	0.0014	0.0094	1.16E-04	713.7360		
Perfluorononanoic acid	0.0013	0.0088	8.33E-03	368.6974		
Perfluorohexanoic acid	0.0076	0.0507	2.33E-04	183.6569		
assium perfluorobutanesulfonate	0.0087	0.0581	2.75E-02	101.5252		
Perfluorooctanesulfonic acid	0.0073	0.0490	5.38E-02	57.1902		
rfluoro(4-methoxybutanoic) acid	0.0142	0.0950	2.97E-01	26.7545		
Perfluorobutanoic acid	0.1032	0.6927	1.68E-05	19.8299		
2H,3H,3H-Perfluorooctanoic acid	0.0072	0.0483	5.15E-01	15.2577		H₃C、
fluoro-3,6,9-trioxatridecanoic acid	0.0026	0.0176	6.38E-01	7.9748		1.30
2 Fluorotelomer sulfonic acid	0.0142	0.0951	5.55E+00	1.5874		I
Ethylperfluorooctanesulfonamide	0.0464	0.3110	5.57E+00	0.9485	Π	0=
1ethylperfluorooctanesulfonamide	0.0113	0.0757	7.43E+00	0.7633	<b>))</b>	F
Perfluorooctanesulfonamide	0.0229	0.1536	3.60E+01	0.1630	<b>–</b>	
	assium perfluorohexanesulfonate Ammonium perfluorooctanoate Perfluorononanoic acid Perfluorohexanoic acid assium perfluorobutanesulfonate Perfluorooctanesulfonic acid rfluoro(4-methoxybutanoic) acid Perfluorobutanoic acid 2H,3H,3H-Perfluorooctanoic acid fluoro-3,6,9-trioxatridecanoic acid 1:2 Fluorotelomer sulfonic acid Ethylperfluorooctanesulfonamide	assium perfluorohexanesulfonate0.0011Ammonium perfluorooctanoate0.0014Perfluorononanoic acid0.0013Perfluorohexanoic acid0.0076assium perfluorobutanesulfonate0.0087Perfluorooctanesulfonic acid0.0073rfluoro(4-methoxybutanoic) acid0.0142Perfluorobutanesulfonic acid0.0142Perfluorobutanoic acid0.0142Perfluorobutanoic acid0.0142Perfluorobutanoic acid0.00722H,3H,3H-Perfluorooctanoic acid0.0026L2 Fluorotelomer sulfonic acid0.0142Ethylperfluorooctanesulfonamide0.0464Iethylperfluorooctanesulfonamide0.0113	assium perfluorohexanesulfonate0.00110.0075ammonium perfluorooctanoate0.00140.0094Perfluorononanoic acid0.00130.0088Perfluorohexanoic acid0.00760.0507assium perfluorobutanesulfonate0.00870.0581Perfluorooctanesulfonic acid0.00730.0490rfluoro(4-methoxybutanoic) acid0.01420.0950Perfluorobutanesulfonic acid0.10320.69272H,3H,3H-Perfluorooctanoic acid0.00720.0483fluoro-3,6,9-trioxatridecanoic acid0.01420.0951Ethylperfluorooctanesulfonamide0.04640.3110Iethylperfluorooctanesulfonamide0.01130.0757	assium perfluorohexanesulfonate 0.0011 0.0075 3.82E-07   Ammonium perfluorooctanoate 0.0014 0.0094 1.16E-04   Perfluorononanoic acid 0.0013 0.0088 8.33E-03   Perfluorohexanoic acid 0.0076 0.0507 2.33E-04   assium perfluorobutanesulfonate 0.0087 0.0581 2.75E-02   Perfluorooctanesulfonic acid 0.0073 0.0490 5.38E-02   rfluoro(4-methoxybutanoic) acid 0.0142 0.0950 2.97E-01   Perfluorobutanoic acid 0.1032 0.6927 1.68E-05   2H,3H,3H-Perfluorooctanoic acid 0.0072 0.0483 5.15E-01   fluoro-3,6,9-trioxatridecanoic acid 0.0026 0.0176 6.38E-01   :2 Fluorotelomer sulfonic acid 0.0142 0.0951 5.55E+00   Ethylperfluorooctanesulfonamide 0.0464 0.3110 5.57E+00	assium perfluorohexanesulfonate0.00110.00753.82E-07894.5132ammonium perfluorooctanoate0.00140.00941.16E-04713.7360Perfluorononanoic acid0.00130.00888.33E-03368.6974Perfluorohexanoic acid0.00760.05072.33E-04183.6569assium perfluorobutanesulfonate0.00870.05812.75E-02101.5252Perfluorooctanesulfonic acid0.00730.04905.38E-0257.1902rfluoro(4-methoxybutanoic) acid0.01420.09502.97E-0126.7545Perfluorobutanesulfonic acid0.00720.04835.15E-0115.2577fluoro-3,6,9-trioxatridecanoic acid0.00260.01766.38E-017.9748t:2 Fluorotelomer sulfonic acid0.04640.31105.57E+001.5874Ethylperfluorooctanesulfonamide0.01130.07577.43E+000.7633	assium perfluorohexanesulfonate 0.0011 0.0075 3.82E-07 894.5132   Ammonium perfluorooctanoate 0.0014 0.0094 1.16E-04 713.7360   Perfluorononanoic acid 0.0013 0.0088 8.33E-03 368.6974   Perfluorohexanoic acid 0.0076 0.0507 2.33E-04 183.6569   assium perfluorobutanesulfonate 0.0087 0.0581 2.75E-02 101.5252   Perfluorooctanesulfonic acid 0.0073 0.0490 5.38E-02 57.1902   rfluoro(4-methoxybutanoic) acid 0.1032 0.6927 1.68E-05 19.8299   2H,3H,3H-Perfluorooctanoic acid 0.0072 0.0483 5.15E-01 15.2577   Ruoro-3,6,9-trioxatridecanoic acid 0.0026 0.0176 6.38E-01 7.9748   Ethylperfluorooctanesulfonamide 0.0464 0.3110 5.57E+00 1.5874   Ethylperfluorooctanesulfonamide 0.0113 0.0757 7.43E+00 0.7633

34



## What is the big picture of this PFAS toxicity effort?





## **Summary of findings**

- Experimental *in vitro* toxicokinetic data (F<sub>u</sub> and Cl<sub>hepatic</sub>) are being measured on over 120 PFAS for use in IVIVE modeling
- Multi-residual LC-MS/MS method developed to analyze more than 60 unique PFAS
- Plasma protein binding data indicate high binding rates, with 75% exhibiting  $F_u$  values from 0.001 0.05
- Assuming an external exposure of 1  $\mu$ g/kg/day, C<sub>ss</sub> predictions ranged from 0.16-895  $\mu$ M, with a median value of 23.29  $\mu$ M
- These C<sub>ss</sub> estimates eventually will be combined with other highthroughput screening data to help identify PFAS risk to humans
- Continuing data generation for additional PFAS and toxicokinetic assays for bioavailability, metabolite identification, and renal reuptake



## **Acknowledgements**

#### ORD-CCTE

- Lucas Albrecht
- Mike DeVito
- Annette Guiseppi-Elie
- Josh Harrill
- Keith Houck
- Mike Hughes
- Richard Judson
- Jen Korol-Bexell
- Anna Kreutz
- Stephanie Padilla
- Grace Patlewicz
- Matthew Phillips
- Ann Richard
- Tim Shafer
- Adam Swank

- Rusty Thomas
- John Wambaugh
- Barbara Wetmore
- Antony Williams

#### ORD-CEMM

- Scott Clifton
- Matt Henderson
- James McCord
- Larry McMillan
- Mark Strynar
- NTP
  - David Crizer
- Waters
  - Aurelie Marcotte
  - Kari Organtini





#### smeltz.marci@epa.gov