

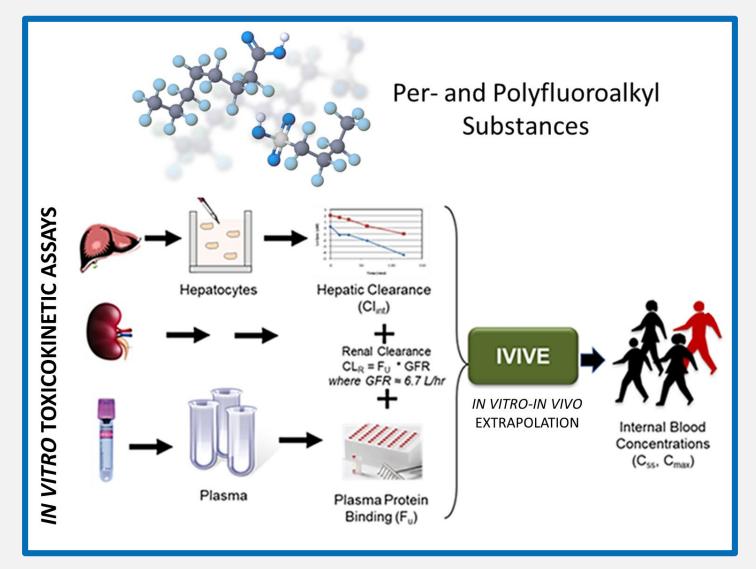
# Evaluating Per- and Polyfluoroalkyl Substances (PFAS) by In Vitro Toxicokinetic Data Generation with In Vitro-In Vivo Extrapolation (IVIVE)

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U.S. Environmental Protection Agency, Office of Research and Development, Center for Computational Toxicology and Exposure

Presentation for ASCCT 9th Annual Meeting





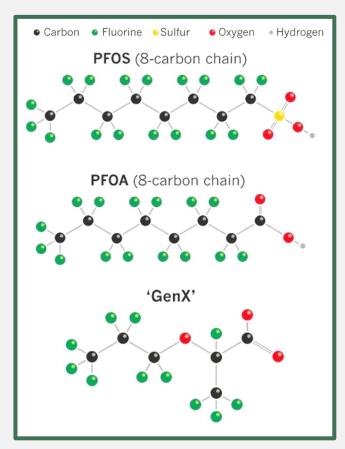


## What are PFAS?

- PFAS = per- and polyfluoroalkyl substances
- Man-made chemicals used in industry and consumer products worldwide since the 1950s
- Repel water, resist heat, and protect surfaces









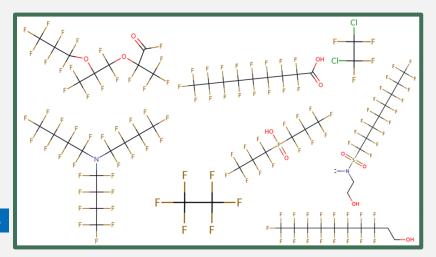
## Why are PFAS receiving lots of attention?

## Widespread occurrence

- PFAS in 97% of American population
- Even in arctic polar bears

### Persistence

- Carbon-fluorine bonds are some of the strongest
- Little degradation in environment

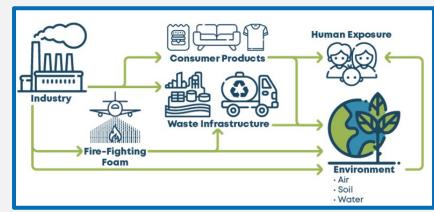


### Bioaccumulative

- Accumulate over time
- Absorption > elimination

### Abundance

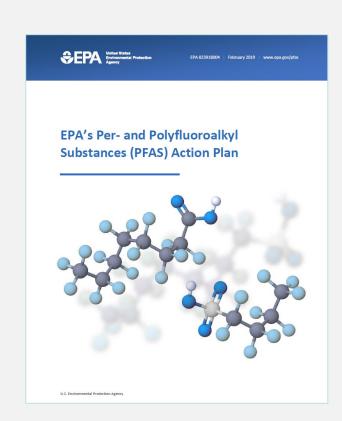
- 6,648 PFAS-like structures on EPA's Dashboard
- 1,223 PFAS on TSCA inventory with 602 currently in use in USA





## What is the EPA doing about PFAS?

- EPA PFAS Action Plan (2019)
  - Assist states, tribes, and communities address PFAS with short-term solutions and long-term strategies to address PFAS
- PFAS-Related Challenges
  - Developing/validating laboratory analytical methods for measuring PFAS
  - Assessing PFAS chemical toxicity
  - Developing standard toxicity values for PFAS chemicals
  - Characterizing potential human exposure pathways
  - Managing PFAS containing materials and waste
  - Testing drinking water treatment technologies
  - Identifying site remediation technologies





## Which PFAS are we interested in?

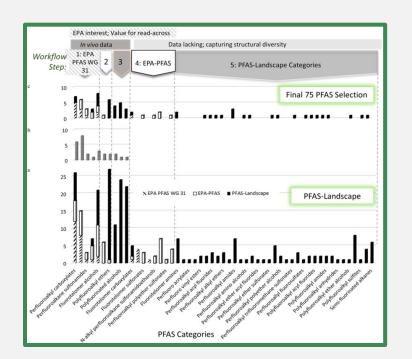
- PFAS Screening Library creation: PFAS Landscape
  - Maximize read-across
  - Capture structural diversity

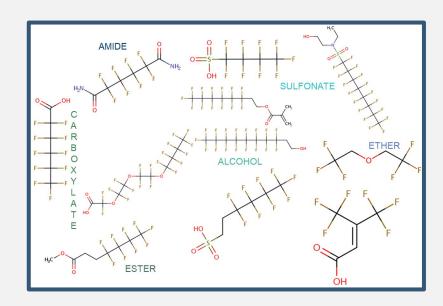
- Brief Communication

  A Chemical Category-Based Prioritization Approach for Selecting 75 Per- and Polyfluoroalkyl Substances (PFAS) for Tiered Toxicity and Toxicokinetic Testing Grace Patlevicz, Ann M. Richand, Antony J. Williams, Christopher M. Grulke, P. Reeder Sams, J. Jason Lambert, Pamela D. Noyes, Michael J. DeVito, \* Ronald N. Hines, \* Mark Strynar, \* Annette Guiseppi-Elie, \* and Russell S. Thomas' Environmental Health Perspectives

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  127(1) January 2019
- Initially, 75 PFAS selected from the PFAS Landscape
- Now, PFAS of interest for testing includes nearly <u>200</u> unique structures with <u>430</u> unique PFAS in the Landscape

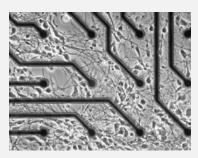






## How are we examining these PFAS?

- A range of targeted and tiered high-throughput toxicity assays to serve as guide for potential human health risk
- New approach methodologies (NAMs) used
  - Alternative test methods and strategies to reduce, refine, and/or replace mammalian animals
  - In vitro tests/assays, in chemico assays, in silico algorithms
- Endpoints for PFAS work
  - Hepatotoxicity
  - Immunotoxicity
  - Developmental toxicity
  - Mitochondrial toxicity
  - In vitro toxicokinetic assays

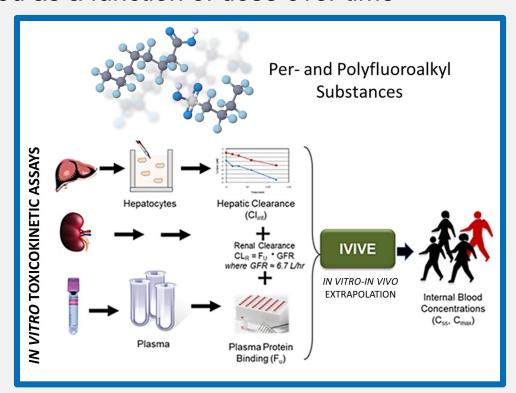






## What are in vitro toxicokinetic assays?

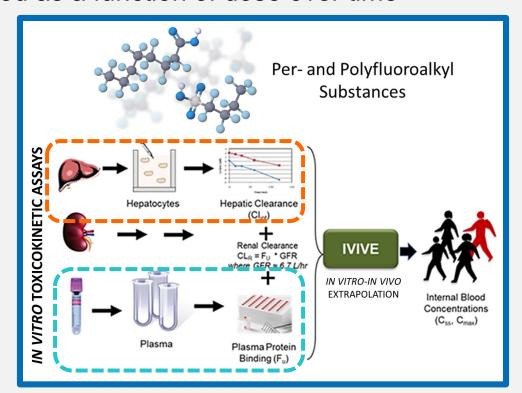
- Toxicokinetics (TK): the study of how a substance gets into the body and what happens to it in the body
  - Can be used to look at how chemicals move throughout the body and lead to harmful effects
  - Often viewed as a function of dose over time





## What are in vitro toxicokinetic assays?

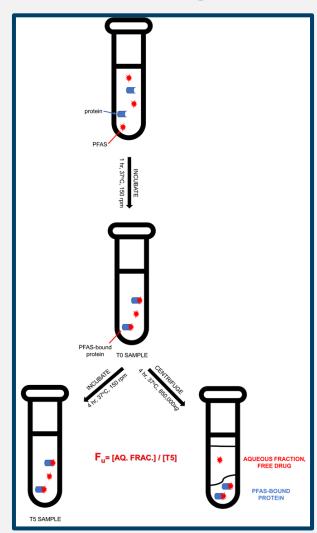
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## What is plasma protein binding?

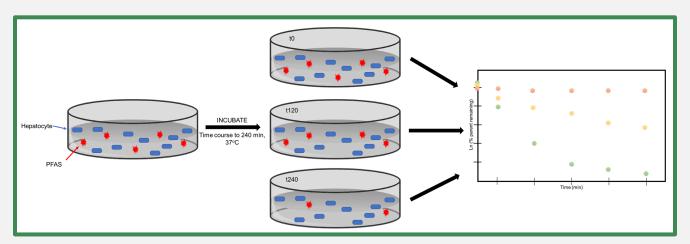
- Assay to assess the free (unbound) fraction of chemical to proteins within the blood
  - F<sub>u</sub>
  - Unbound molecules permeate through cell membranes to reach 'target'
  - Determine by equilibrium dialysis, ultrafiltration, and/or ultracentrifugation
- Ultracentrifugation assay used for PFAS analysis
  - Human plasma (10-donor pool, mixed sex) centrifuged to separate aqueous fraction from albumin, lipoproteins, and fatty acids
  - Mixtures of up to 4 PFAS (10 μM) were included with each plasma sample, run in triplicate





## What is in vitro hepatic clearance?

- Hepatic clearance (CL<sub>H</sub>) is measure of the rate of elimination of a chemical from the liver
- Models to study metabolism include human liver microsomes, recombinantly expressed enzymes, and hepatocytes
- Substrate depletion approach utilized for PFAS work
  - Primary human hepatocytes (50-donor pool, mixed sex) at 1 μM PFAS concentration
  - Time course: 0, 15, 30, 60, 90, 120, and 240 min with non-linear regression fit
  - Work completed by collaborator at National Toxicology Program [David Crizer]

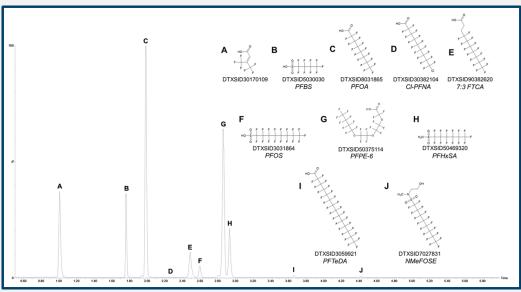




## How do we analyze these assay samples?

- Both assays require concentration determination of parent PFAS
- EPA has a range of analytical capabilities (single quads, triple quads, high resolution mass specs)
  - Ultra-high-performance liquid chromatography tandem mass spectrometer used

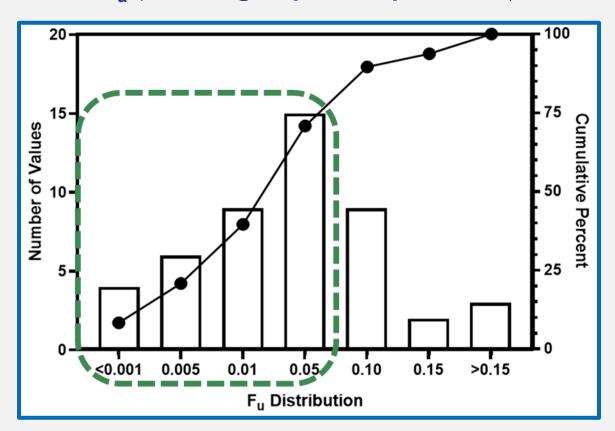




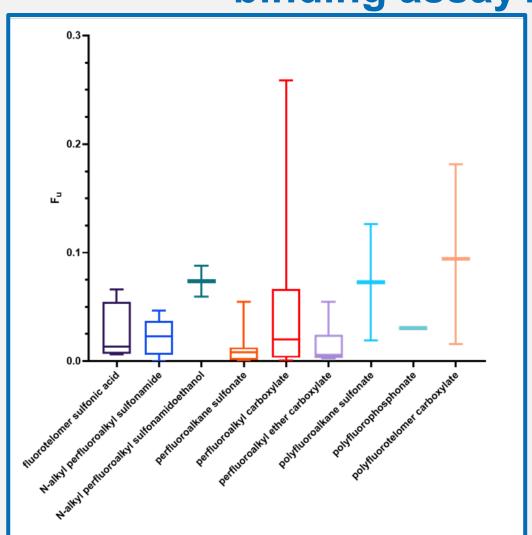


50 LC-able PFAS have determined fraction unbound data

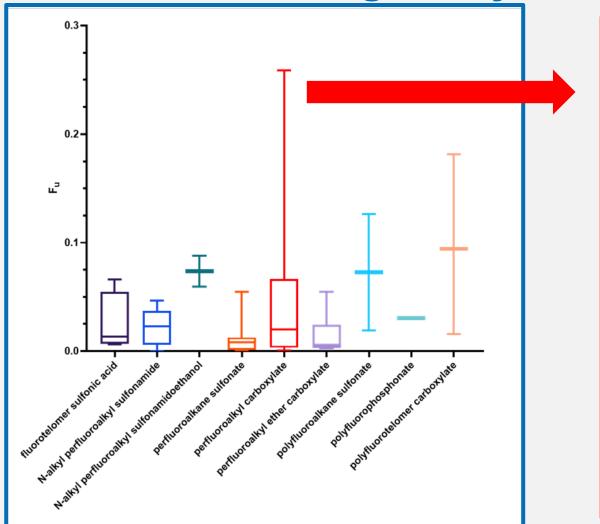
**F**<sub>...</sub> ↓ binding to plasma proteins ↑

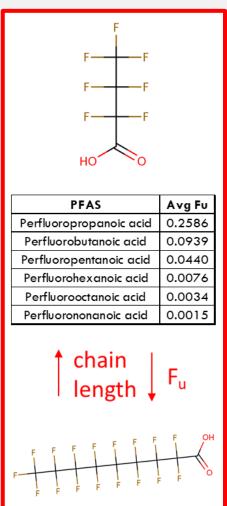




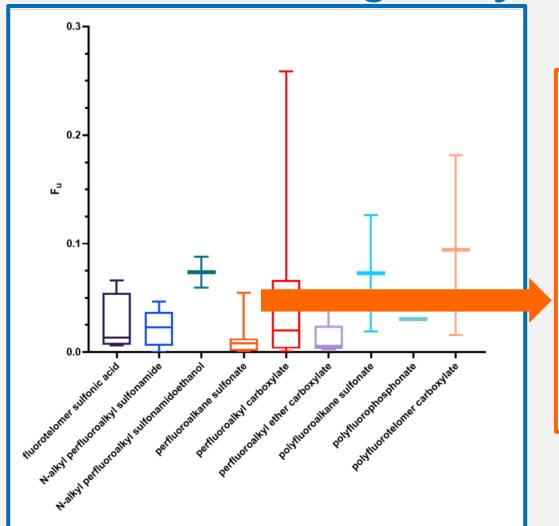


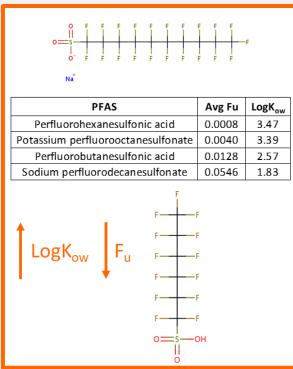








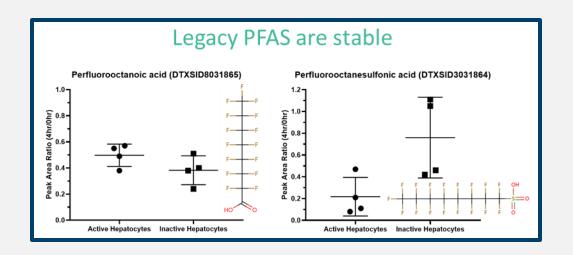






## Any observations from the hepatic clearance assay?

- More than 20 LC-able PFAS assessed
- 1. In vitro hepatic clearance screen
  - 0 and 4 hr time points for active and inactive hepatocytes
  - Compared time ratios to examine for clearance potential
  - Ratio of 1 indicates no loss over time





## Any observations from the hepatic clearance assay?

- More than 20 LC-able PFAS assessed
- 2. Metabolic stability time course
  - 0, 0.25, 0.50, 1, 1.5, 2, 4 hr time points
  - Non-linear fit to determine half-life (T<sub>1/2</sub>)

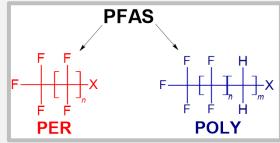
Compound Name	Half-life (min)	Clearance (µL/min/million cells)				
Perfluorobutanoic acid	44769343	1.55E-05				
Potassium perfluorohexanesulfonate	21340366	3.25E-05				
Perfluorohexanoic acid	237257	2.92E-03				
Ammonium perfluorooctanoate	88735	7.81E-03				
Potassium perfluorobutanesulfonate	2300	3.01E-01				
Perfluorononanoic acid	1155	6.00E-01				
Perfluorooctanesulfonic acid	990	7.00E-01				
Perfluoro(4-methoxybutanoic) acid	346.5	2.00E+00				
2H,2H,3H,3H-Perfluorooctanoic acid	101.4	6.83E+00				
N-Ethylperfluorooctanesulfonamide	57	1.22E+01				
3-(Perfluoro-2-butyl)propane-1,2-diol	35.87	1.93E+01				
Perfluoro-3,6,9-trioxatridecanoic acid	29.71	2.33E+01				
Nonafluoropentanamide	25.45	2.72E+01				
3,3-Bis(trifluoromethyl)-2-propenoic acid	19.77	3.51E+01				
4:2 Fluorotelomer sulfonic acid	17.5	3.96E+01				
Octafluoroadipamide	12.8	5.41E+01				
Perfluoropentanamide	10.63	6.52E+01				
N-Methylperfluorooctanesulfonamide	10.17	6.81E+01				
2,2,3,3,4,4-Hexafluorobutanoic acid	4.209	1.65E+02				
Perfluorooctanesulfonamide	2.789	2.48E+02				

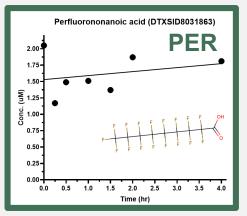


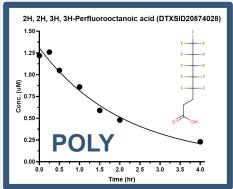
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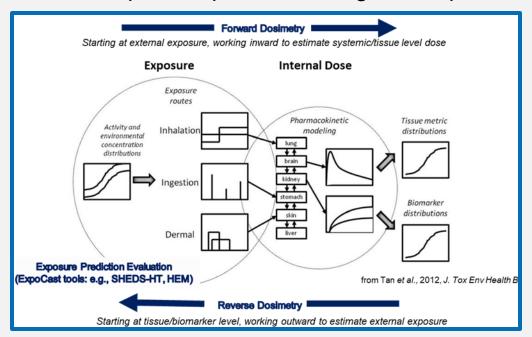






## What is IVIVE?

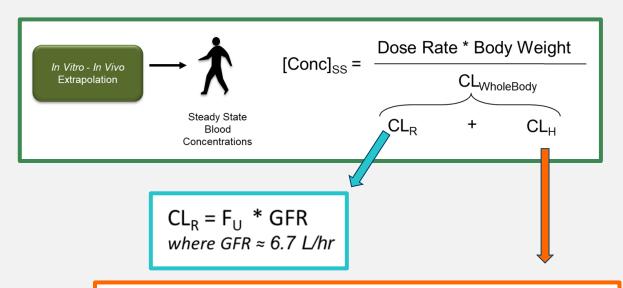
- In vitro-in vivo extrapolation = IVIVE
  - Model approach that allows in vitro data to be extrapolated to estimate corresponding in vivo effects
  - Start at tissue/biomarker level → estimate external exposure
- Steady-state concentration (C<sub>ss</sub>)
  - Concentration of compound in body that stays consistent
  - This takes into account plasma protein binding and hepatic clearance data





## Per- and Polyfluoroalkyl Substances Per- and Polyfluoroalkyl Substances Plasma Perotein Binding (F, )

## What is IVIVE?



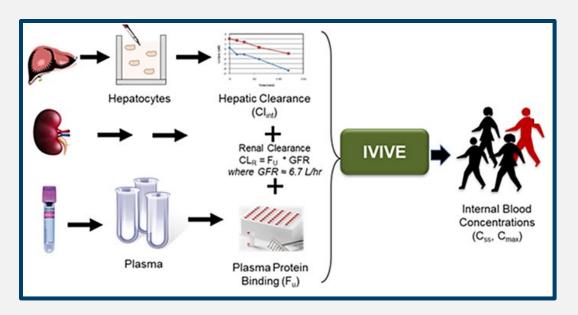
## 

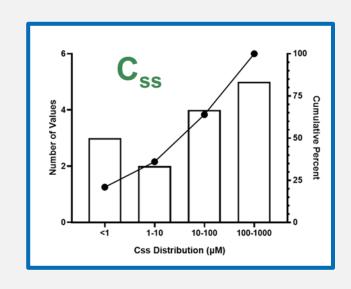
## **Assumptions**

Exposure at 1 µg/kg/day
Linear kinetics
100% oral bioavailability

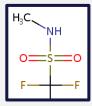


## What did IVIVE show with PFAS data?





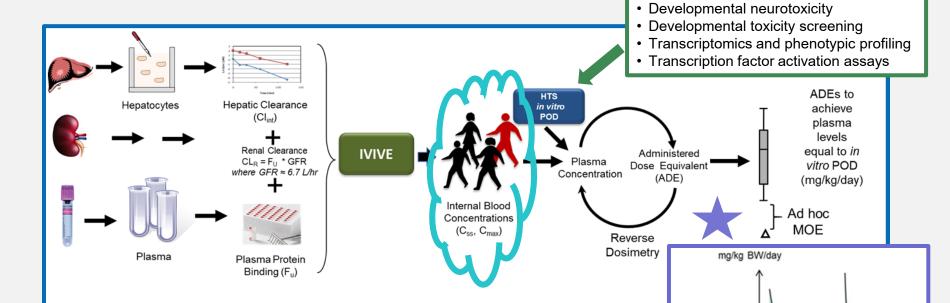
	Compound Name	Fu	CI <sub>renal</sub> (L/hr)	Cl <sub>hepatic</sub> (L/hr)	Css (μM)	
	Potassium perfluorohexanesulfonate	0.0011	0.0075	3.82E-07	894.5132	1
	Ammonium perfluorooctanoate	0.0014	0.0094	1.16E-04	713.7360	١١
	Perfluorononanoic acid	0.0013	0.0088	8.33E-03	368.6974	
	Perfluorohexanoic acid	0.0076	0.0507	2.33E-04	183.6569	
	Potassium perfluorobutanesulfonate	0.0087	0.0581	2.75E-02	101.5252	
	Perfluorooctanesulfonic acid	0.0073	0.0490	5.38E-02	57.1902	1
	Perfluoro(4-methoxybutanoic) acid	0.0142	0.0950	2.97E-01	26.7545	
<b>(</b>	Perfluorobutanoic acid	0.1032	0.6927	1.68E-05	19.8299	
	2H,2H,3H,3H-Perfluorooctanoic acid	0.0072	0.0483	5.15E-01	15.2577	
	Perfluoro-3,6,9-trioxatridecanoic acid	0.0026	0.0176	6.38E-01	7.9748	
	4:2 Fluorotelomer sulfonic acid	0.0142	0.0951	5.55E+00	1.5874	
	N-Ethylperfluorooctanesulfonamide	0.0464	0.3110	5.57E+00	0.9485	7
	N-Methylperfluorooctanesulfonamide	0.0113	0.0757	7.43E+00	0.7633	
	Perfluorooctanesulfonamide	0.0229	0.1536	3.60E+01	0.1630	4
						-





What is the big picture of this PFAS toxicity effort?

High Throughput Screening Assays



Potential Hazard from in vitro with Reverse

**Toxicokinetics** 

Potential Exposure

Ring et al 2017 Environ Int

Higher

Risk

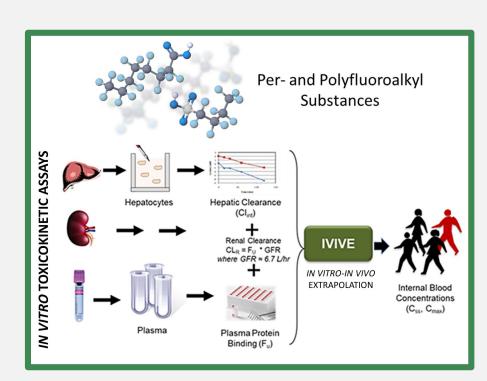
Medium Risk

Generated PFAS data along with human exposure information will assist in informing human health risk assessment and subsequent testing



## **Summary of Findings**

- Experimental in vitro toxicokinetic data (F<sub>u</sub> and Cl<sub>hepatic</sub>) are being measured on over 120 PFAS for use in IVIVE modeling
  - Plasma protein binding data indicate high binding rates, with 75% exhibiting F<sub>u</sub> values from 0.001 – 0.05
  - Assuming an external exposure of 1
     μg/kg/day, C<sub>ss</sub> predictions ranged from 0.16895 μM, with a median value of 23.29 μM
- These C<sub>ss</sub> estimates eventually will be combined with other high-throughput screening data to help identify PFAS risk to humans
- Continuing data generation for additional PFAS and toxicokinetic assays for bioavailability, metabolite identification, and renal reuptake





## **Acknowledgements**

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