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Introduction

- Traditional targeted analysis requires standards for methods development
- Targeted analysis using standards facilitates robust compound quantitation
- NTA studies can acquire standards for confirmation and post-hoc quantitation
- Post-hoc analyte quantitation is subject to increased estimation error
- True quantitative NTA (qNTA) does not utilize structure-matched standards
- qNTA relies on calibration information from one or more surrogate analytes
- Estimation error is larger with qNTA than with post-hoc quantitative analysis
- Strategies are needed to estimate and minimize gNTA estimation error



Concept



- Every HRMS measurement of every compound yields an empirical response factor (RF=abundance/concentration)
- The RF is assumed stable when operating within the linear dynamic range
- The RF is never perfectly stable
- An RF from any surrogate can be used to estimate the concentration of any analyte
- A distribution of RFs across many compounds can be used to estimate the uncertainty about individual concentration predictions
- Models that predict ionization efficiency may be able to reduce quantitative estimation uncertainty



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Quantitative Non-Targeted Analysis: From Data to Decisions

Methods

Data

• NTA data were from EPA's Non-Targeted Analysis Collaborative Trial (ENTACT) • All data were collected and processed using semi-automated techniques • Full dataset included 530 chemicals for ESI+ mode and 237 chemicals for ESI- mode • Each chemical in the full dataset was measured at multiple dilutions • A chemical subset was measured at multiple dilutions in multiple samples

Modeling

Inverse concentration prediction was performed using three methods:

• Traditional calibration curve method:

- Only for chemical subset measured in multiple samples
- Performed using log-log regression with 95% prediction intervals
- Bounded response factor (RF) method:
 - Naïve method that does not consider chemical structure
 - Requires non-parametric estimation of RF 2.5th and 97.5th percentiles

• *Ionization efficiency (IE) estimation method:*

- Uses chemical structures and predicted IE values to restrict possible RF values
- Requires data transformations and linear mixed-effects modeling

Evaluation

• Performed hierarchical bootstrap sampling with five-fold cross validation • Upper confidence limit estimates of concentration used for evaluation • Error quotient (EQ) is the upper confidence limit / true concentration

- Calibration curve method:
- Bounded response factor method:
 - 95% of EQs ≤ 152
 - 50% of EQs \leq 37
- Ionization efficiency estimation method:





- Multiple viable methods for estimating uncertainty in qNTA predictions
- The magnitude of estimation error reflects random and betweenchemical effects
- IE prediction models can help reduce estimation error
- Future models must additionally consider extraction & matrix effects

¹ Fisher and Peter et al. Anal. Bioanal. Chem. *Submitted* ² Groff et al. Anal. Bioanal. Chem. *Accepted* ³ McCord et al. Environ. Int. 2022. doi:10.1016/j.envint.2021.107011

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Results

ESI+ Mode Data:

- 95% of EQs \leq 16
- 50% of EQs ≤ 2
- 95% of EQs \leq 60
- 50% of EQs \leq 10

ESI- Mode Data:

- Calibration curve method:
 - 95% of EQs ≤ 8
 - 50% of EQs \leq 2
- Bounded response factor method:
 - 95% of EQs ≤ 128
 - 50% of EQs \leq 10
- Ionization efficiency estimation method:
 - 95% of EQs ≤ 117
 - 50% of EQs \leq 10

Conclusions & Next Steps

- Upper-bound qNTA estimates
- require hazard-based context for
- risk-based interpretation



References