

Quantitative Non-Targeted Analysis: From Data to Decisions

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Why Does EPA Need Measurement Data?

- **Measurement data needed to ensure chemical safety**

- Characterize risk
- Regulate use & disposal
- Manage human & ecological exposures
- Ensure compliance under federal statutes

Toxic Substances Control Act (TSCA) Compliance Monitoring

To protect federal, state, and tribal health and the environment from unreasonable risks of chemicals, EPA monitors chemical substances.

Safe Drinking Water Act (SDWA) Compliance Monitoring

Providing safe drinking water to states, tribes, and public water systems, EPA monitors water samples collected by the tribes monitor Water Act regulations.

Federal Insecticide, Fungicide and Rodenticide Act Compliance Monitoring

The Federal Insecticide, Fungicide and Rodenticide Act (FIFRA) gives EPA the authority to regulate the registration, distribution, sale and use of pesticides. FIFRA applies to all types of pesticides, including:

Resources and
Guidance
Documents

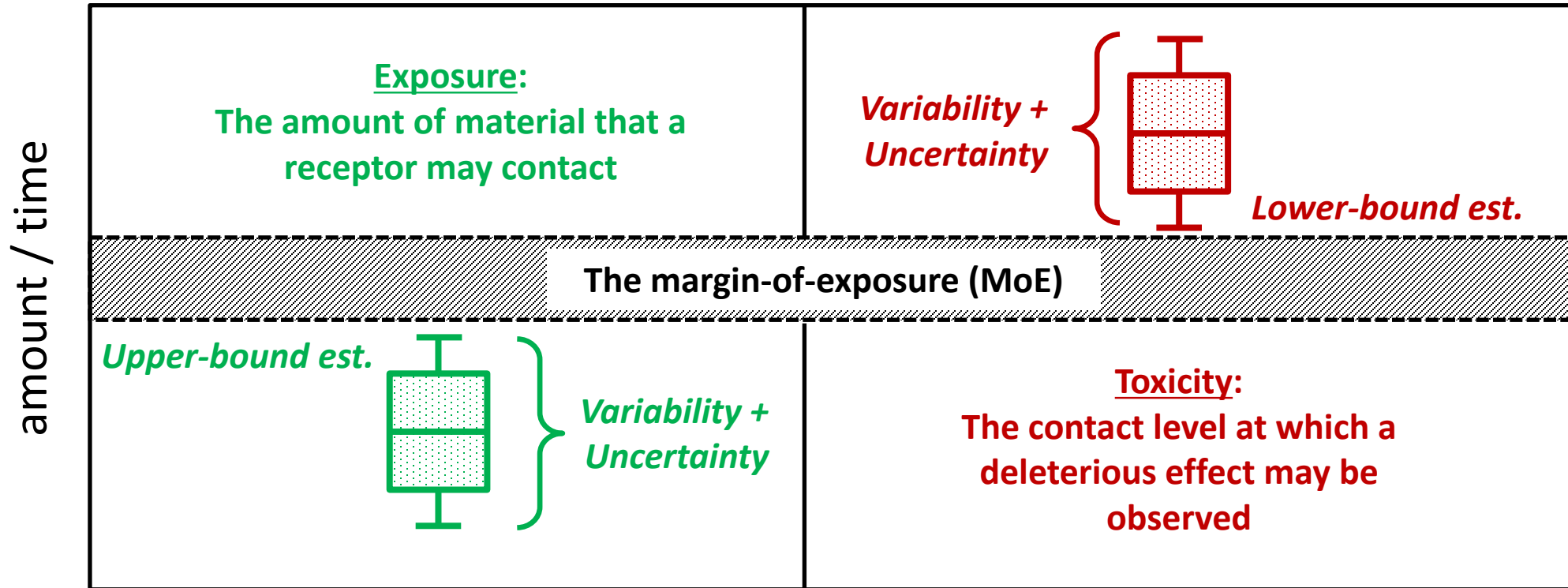
Chemical Monitoring Needs



Challenges

- High-quality monitoring data are unavailable for most chemicals
- Measurement data traditionally generated using “targeted” methods
- Targeted analytical methods:
 - Require *a priori* knowledge of chemicals of interest
 - Produce data for few selected analytes (10s-100s)
 - Require standards for method development & compound quantitation
 - Are blind to emerging contaminants
 - Can’t keep pace with the needs of 21st century risk characterizations
- Quantitative NTA (qNTA) methods must be developed to support provisional risk characterization for emerging contaminants

Risk Characterization Involves Variability and Uncertainty

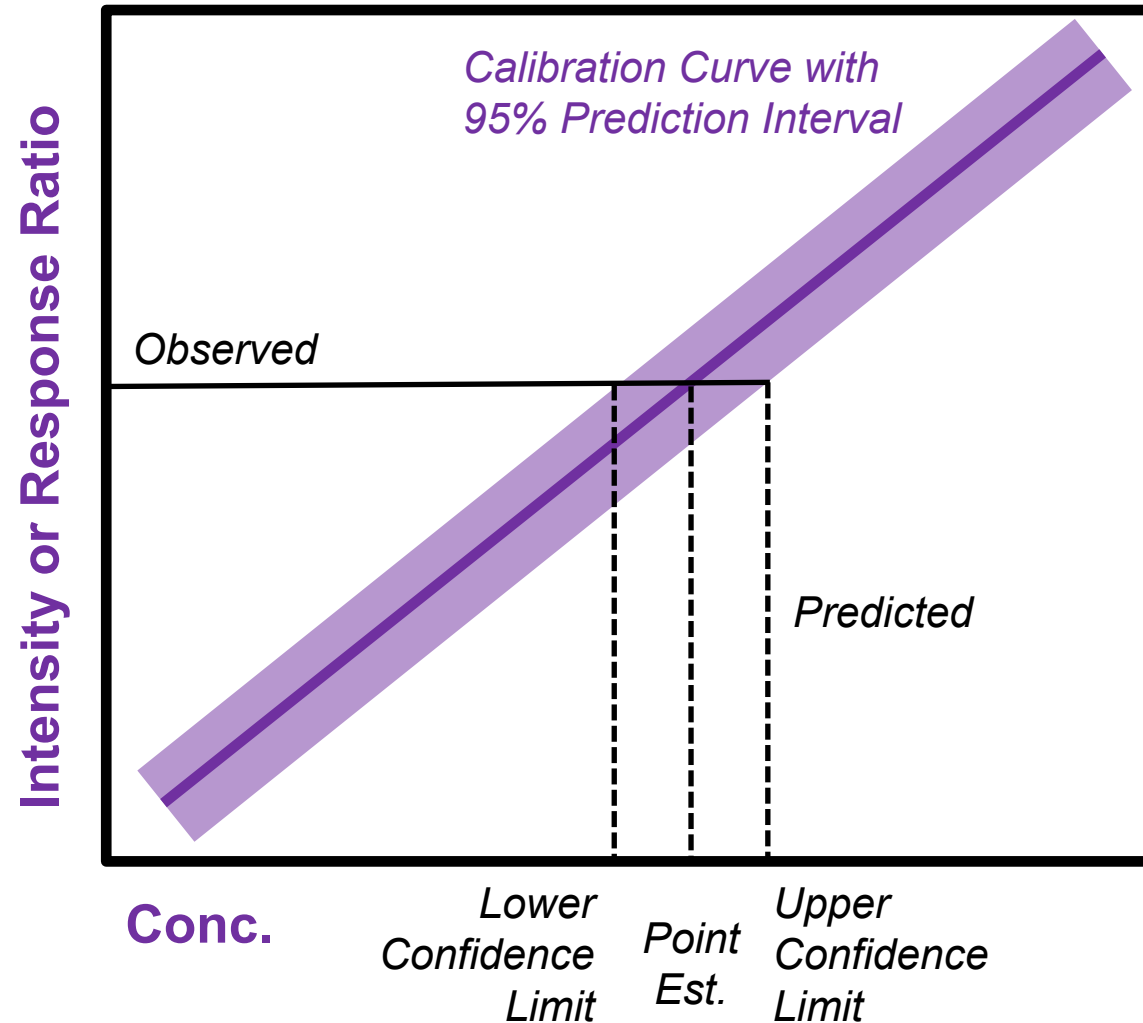


Variability = inherent heterogeneity of phenomena; cannot be reduced, only characterized

Uncertainty = incomplete understanding of phenomena; can be reduced with better methods

Analytical chemistry data help estimate exposure and toxicity, and are both variable and uncertain

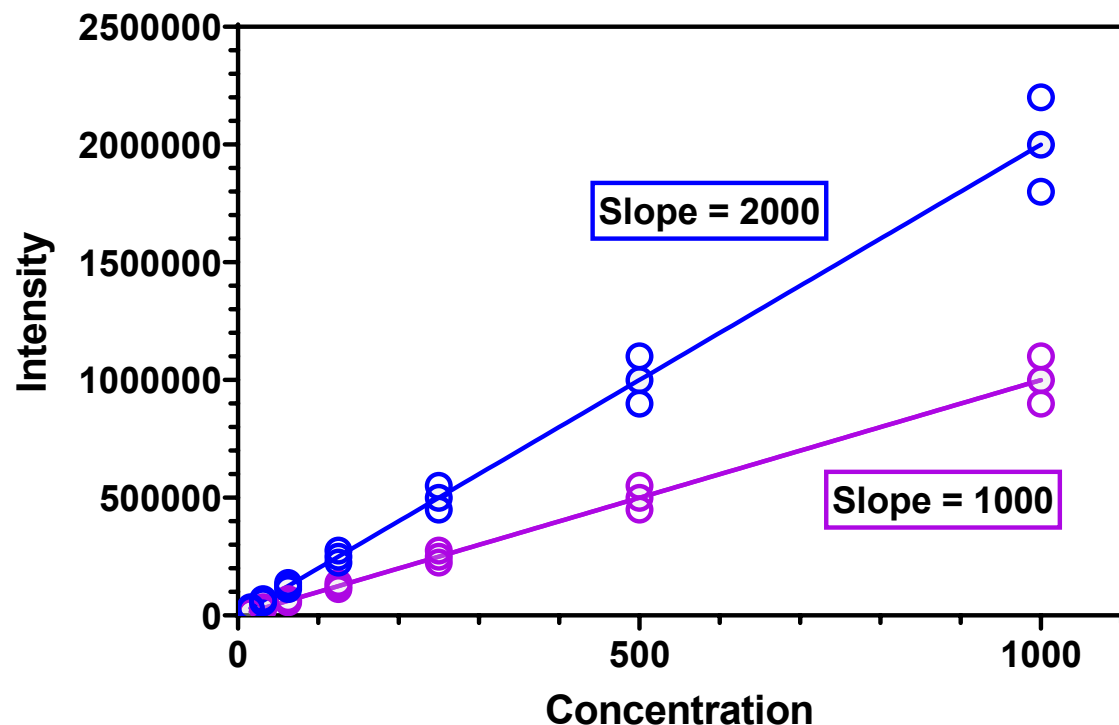
Review of Fundamental Quantitative Method Used with Targeted Analysis



Important Statistical Considerations

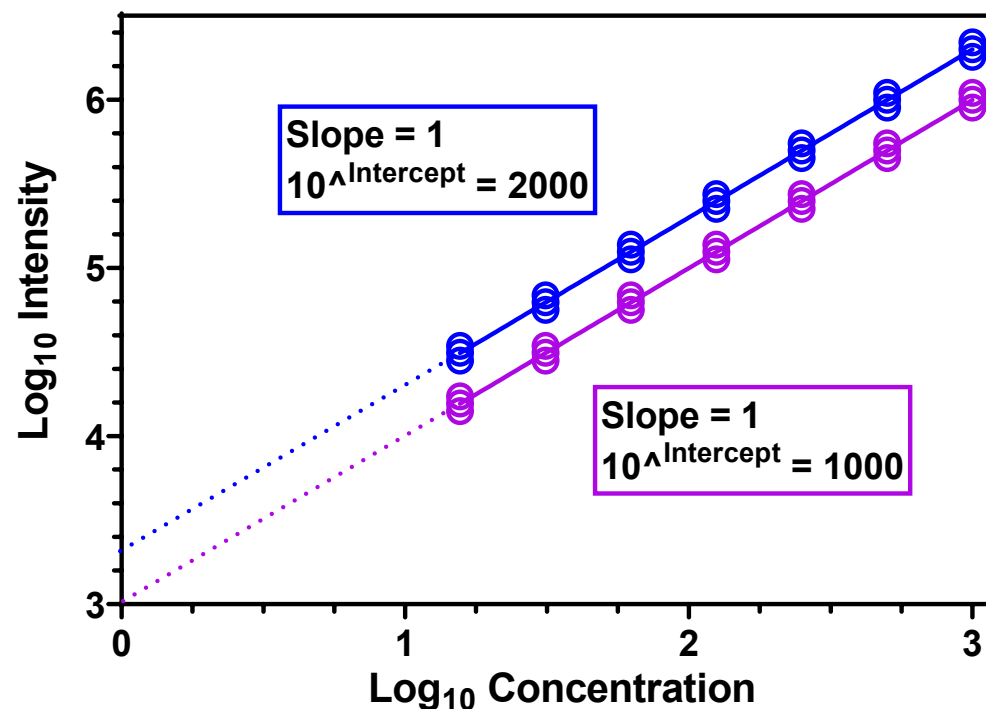
Common Calibration Scenario:

- Unequally spaced dilutions
- Non-uniform measurement variance
- Response Factor (RF) = Intensity/Conc.
- RF = cal. curve slope

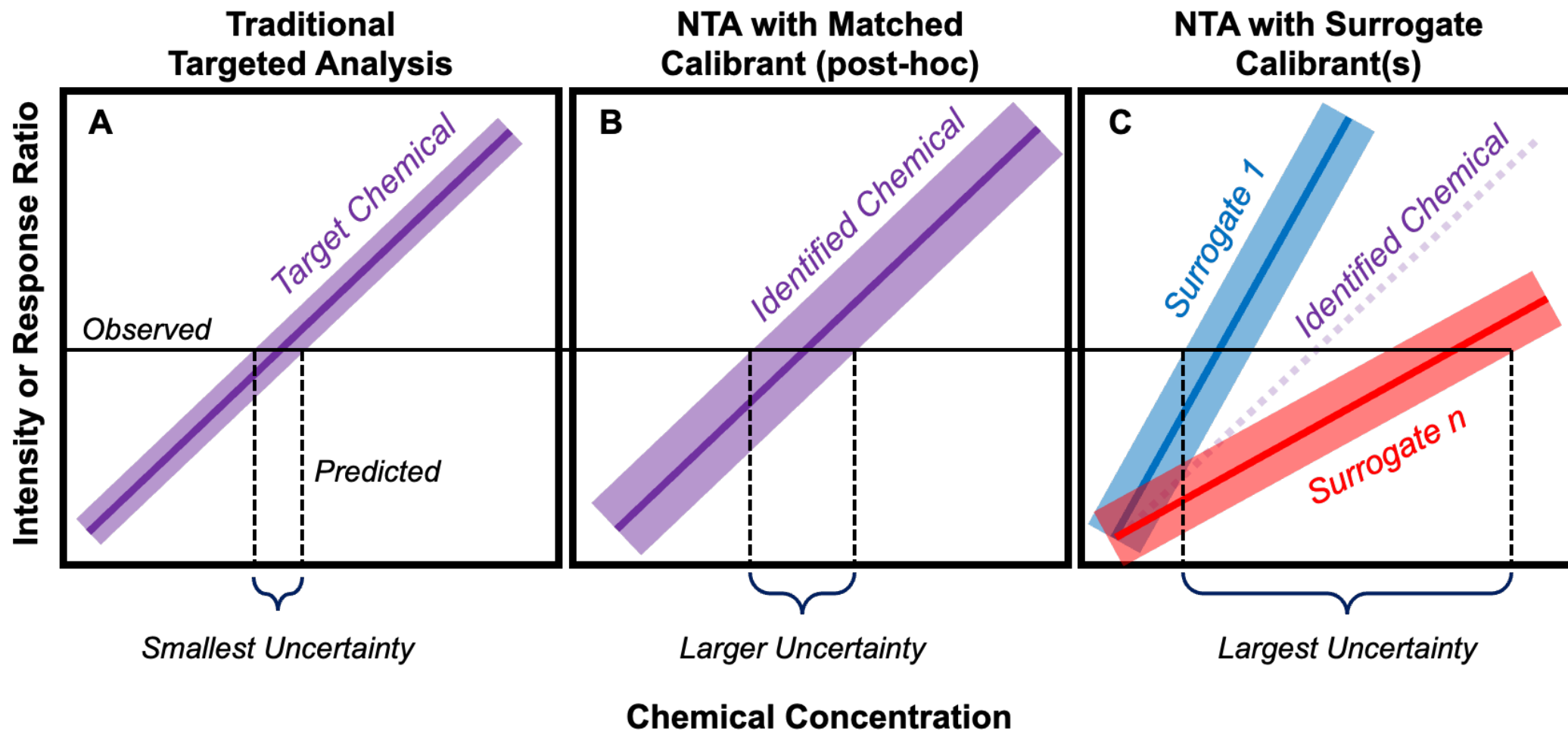


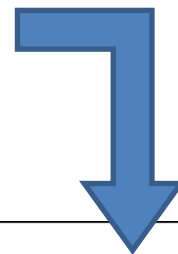
1 Simple Solution → Data Transformation:

- Equally spaced dilutions
- Uniform measurement variance
- Slope ≈ 1 when within linear dynamic range
- RF = $10^{\text{cal. curve intercept}}$



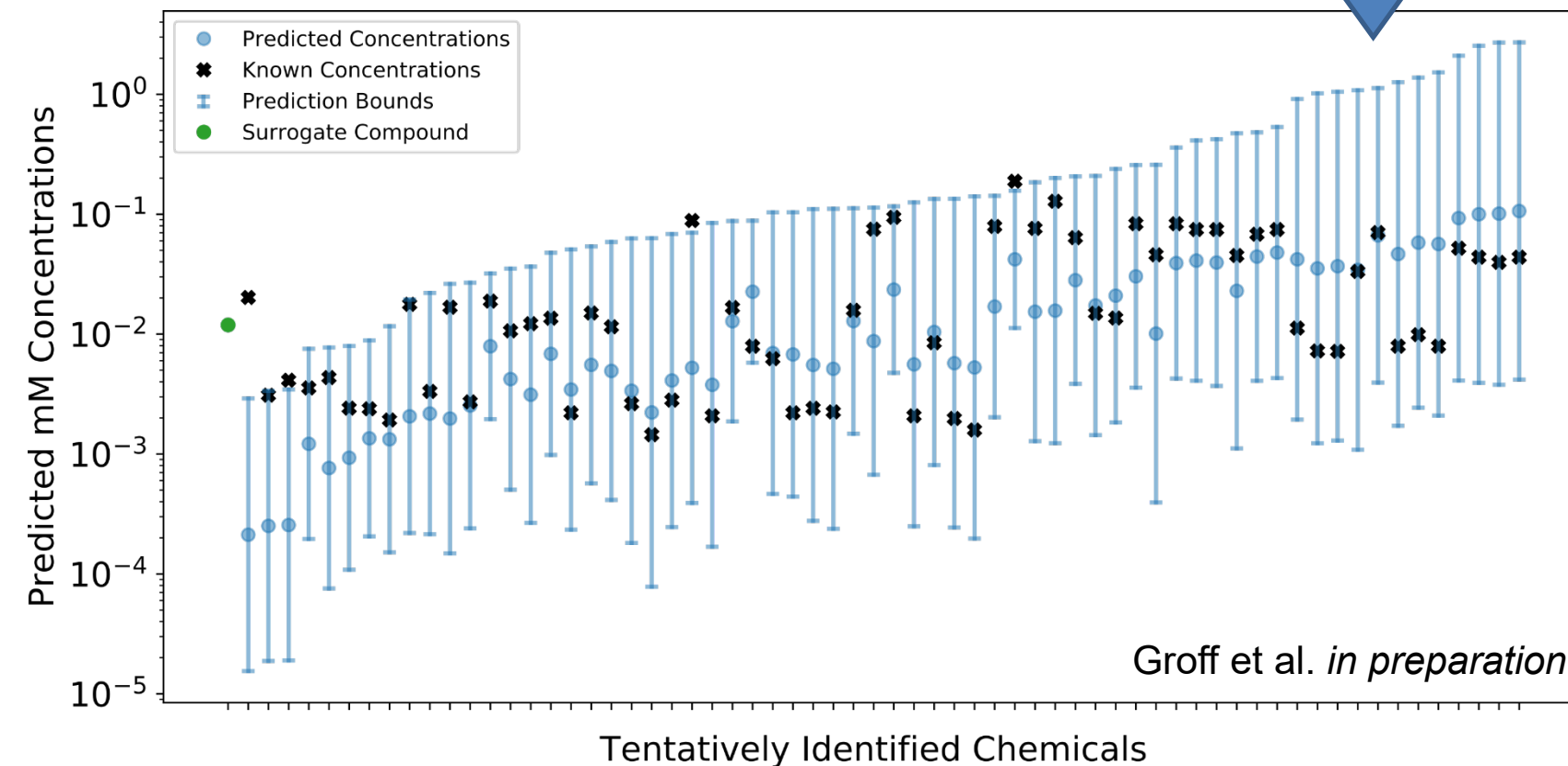
Extension of Fundamental Methods to qNTA





qNTA Proof-of-Concept

- Analysis of Brita filter extracts via GC-HRMS.
- Concentration estimates can be above or below true value.
- Confidence intervals used to bound concentration estimates.
- 95% confidence intervals shown; Can use 99%, 99.9%, etc.
- Tentatively identified compounds ranked by upper-bound estimates.
- Upper-bound estimates compared to level-of-interest to set priorities.
- Priority compounds further examined using targeted methods (when standards can be procured).



The Future of NTA and Chemical Risk Assessment

- The number of labs performing NTA will increase dramatically!
- We're expecting a wealth of NTA data for known (but data-poor) chemicals
 - These data cannot be interpreted using traditional performance metrics
 - **How will risk assessors use new NTA data to support decisions?**
- We're expecting a steady stream of NTA data for newly discovered chemicals
 - Chemical standards won't be readily available (via purchase or synthesis)
 - **How will risk assessors rapidly evaluate the safety of these CECs?**
- **Please visit my poster if you wish to learn more about the development and application of qNTA methods!**

qNTA Research Contributors

The views expressed in this presentation are those of the author(s) and do not necessarily represent the views or policies of the US EPA.



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